

Carbon Capture and Reliable Storage: The Scale of the Challenge

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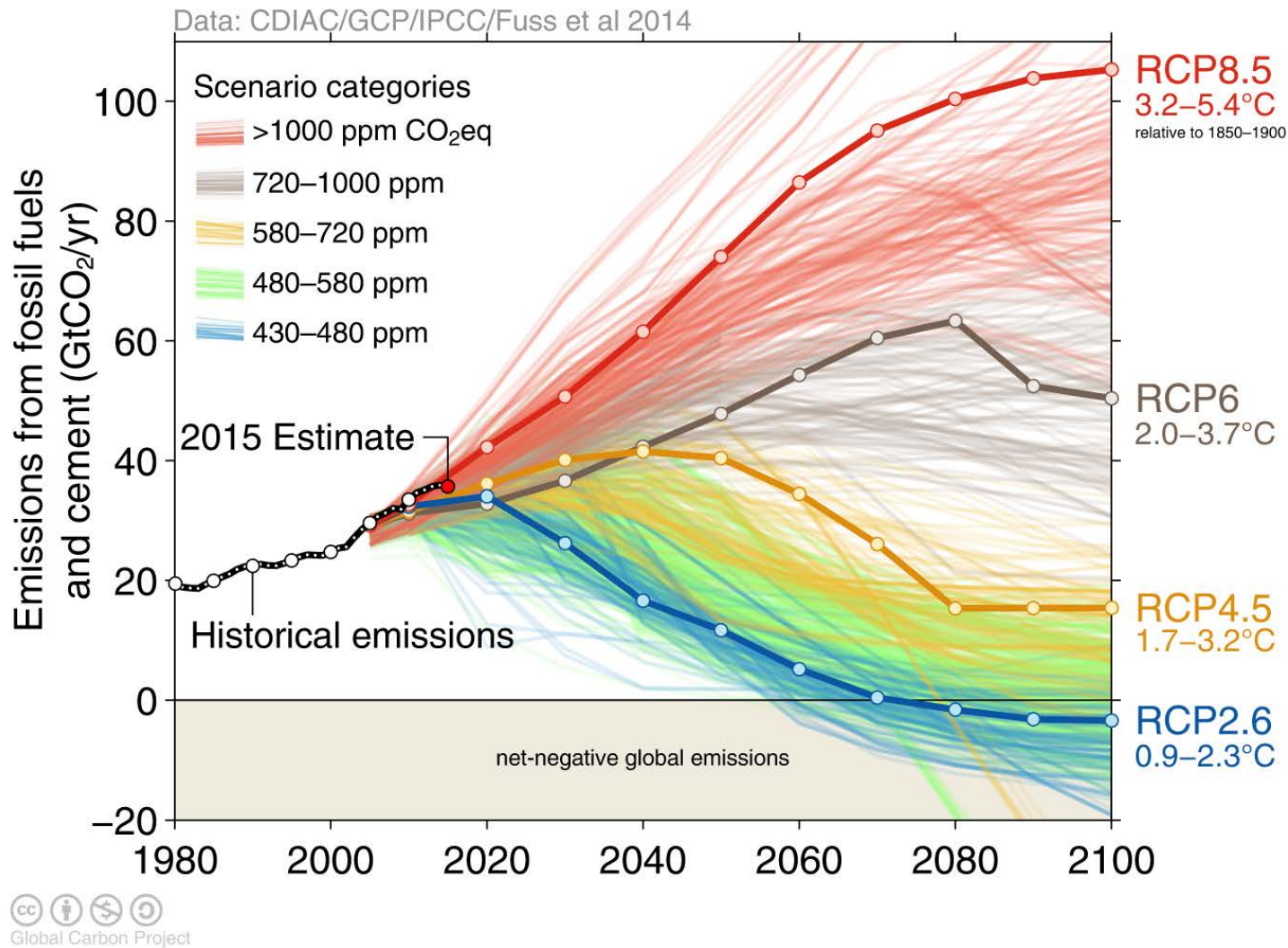
CHEMICAL AND BIOLOGICAL ENGINEERING
COLORADO SCHOOL OF MINES

CALTECH – PLAN B: ENGINEERING A COOLER EARTH
SATURDAY DECEMBER 9TH, 2017



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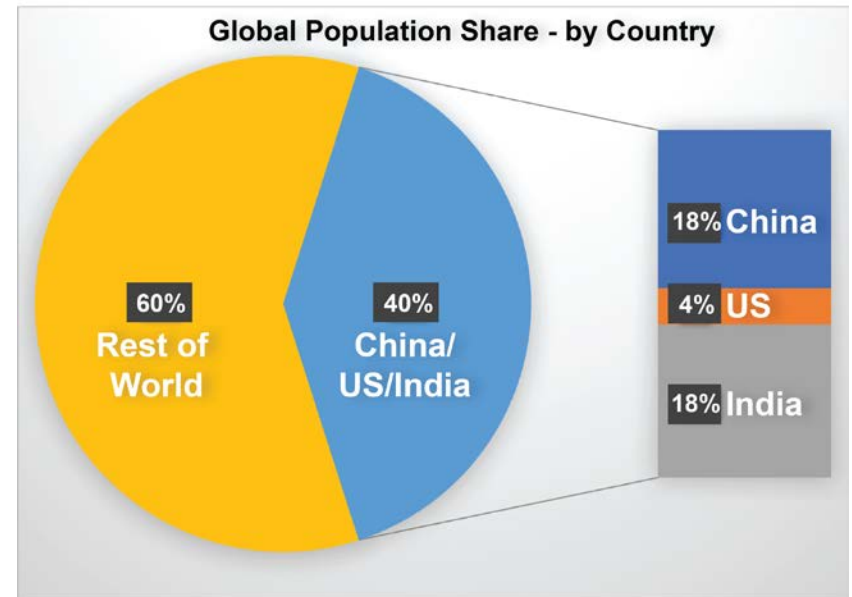
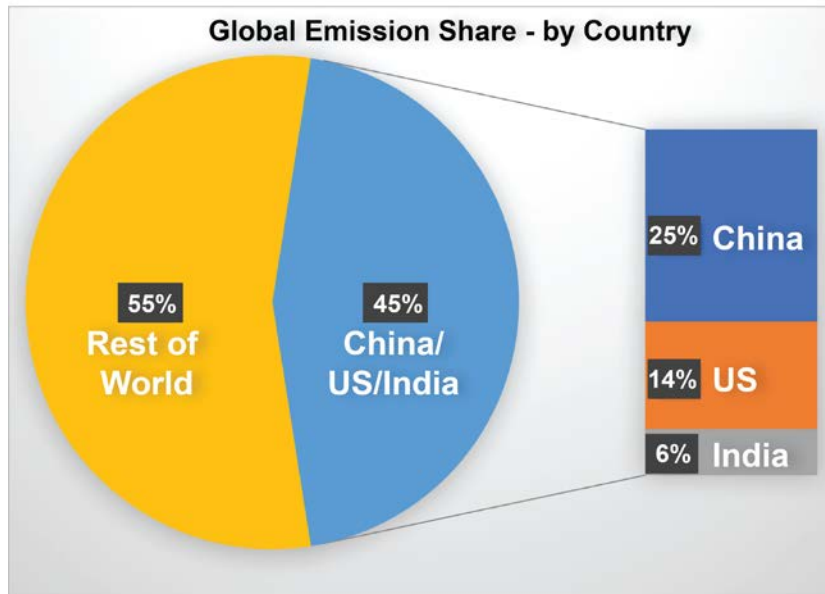
Gigatonne Impacts Are Crucial



Appreciating the Scale of CO₂ Emissions

Fossil Fuels Represent ~ 90% of Global Emissions

Coal (40%), Oil (30%), Natural Gas (20%)



Annual emissions (tCO₂) per capita:

- US ~ 15.6
- China ~ 6.5
- India ~ 1.5



Appreciating the Scale of CO₂ Emissions

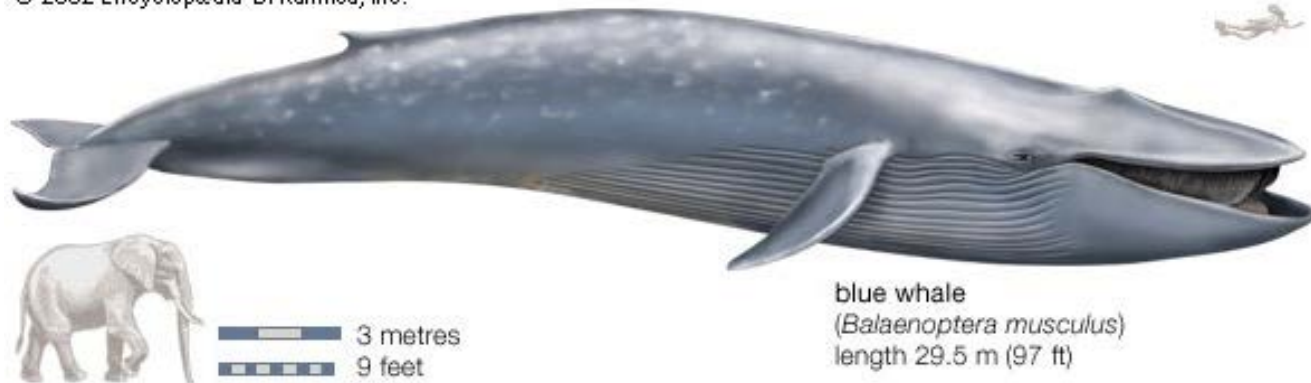
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1. Consider the separation process, adsorption
2. We choose to use an ideal sorbent with today's maximum capacity
3. To capture our per capita emissions requires 150 tonnes of sorbent!

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Message #1

Energetics of Separation Scale with Dilution



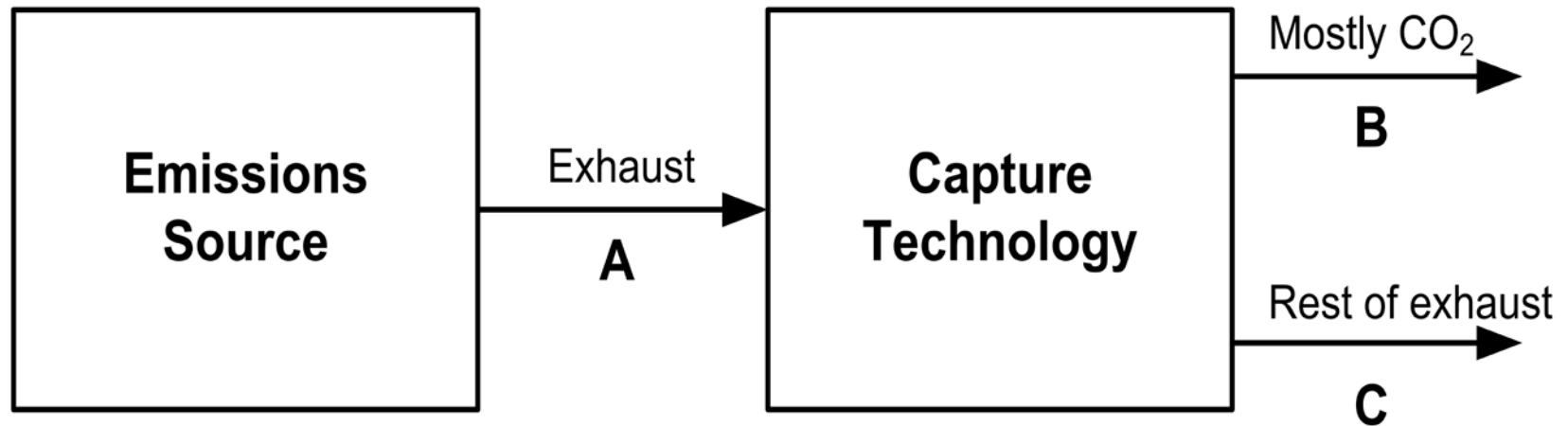
The Majority of the CO₂ Sources are Moderate to Extremely Dilute

Category	% CO ₂ (vol)	Example
High Pressure	varies	Gas Wells (e.g., Sleipner) Synthesis Gas (e.g., IGCC)
High Purity	90-100%	Ethanol Plants Ammonia
Dilute to Moderate	10-15%	Coal-Fired Power Plants → ~ 40% of emissions
Very Dilute	3-7%	Natural Gas Boilers Gas Turbines → ~ 20% of emissions
Extremely Dilute	0.04 – 1%	Ambient Air ^{transport sector} → ~ 25% of emissions Submarines/ Space Craft



Minimum Work for Separation

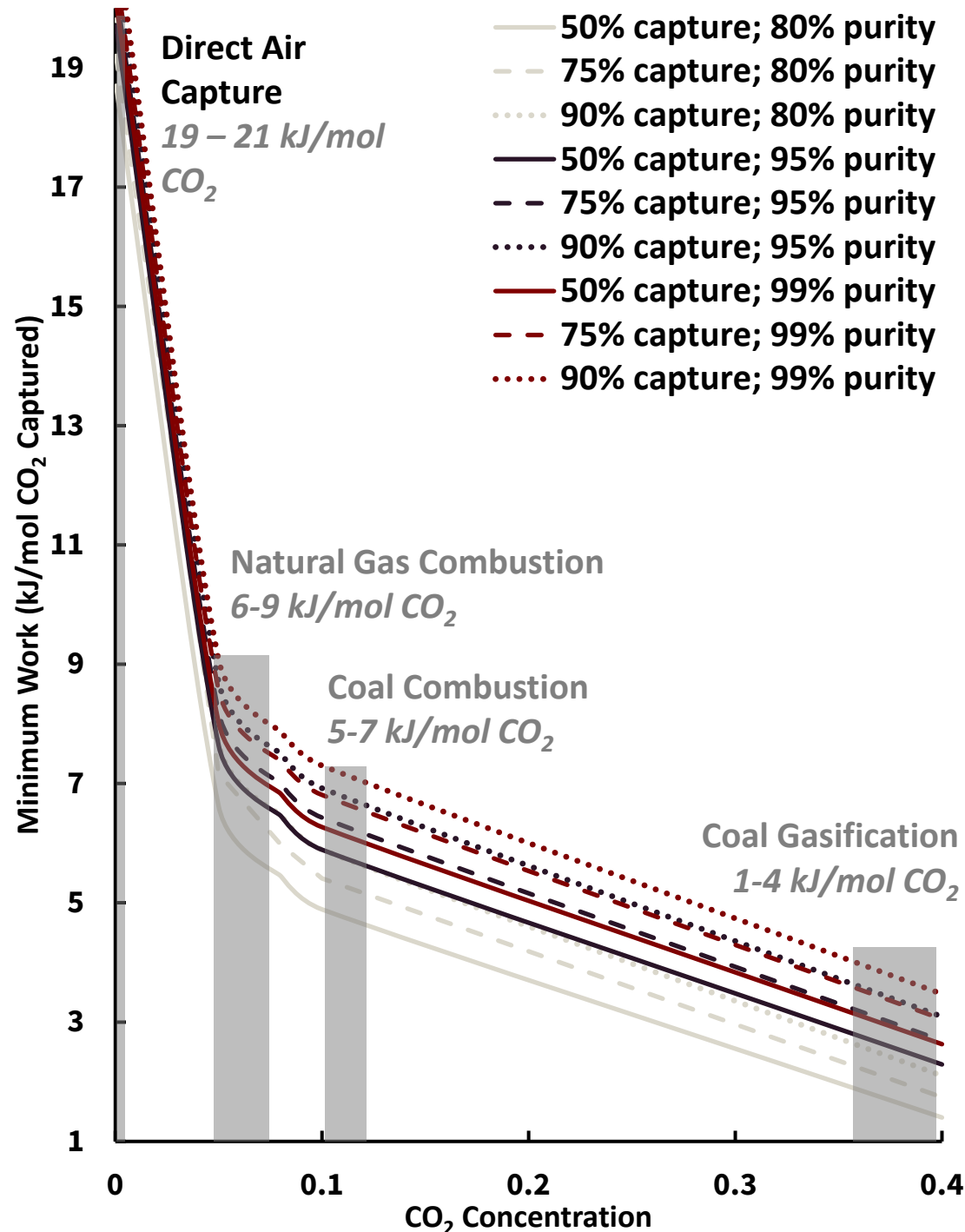
combined first and second laws



$$W_{\min} = RT \left[n_B^{CO_2} \ln(y_B^{CO_2}) + n_B^{B-CO_2} \ln(y_B^{B-CO_2}) \right] + RT \left[n_C^{CO_2} \ln(y_C^{CO_2}) + n_C^{C-CO_2} \ln(y_C^{C-CO_2}) \right] \\ - RT \left[n_A^{CO_2} \ln(y_A^{CO_2}) + n_A^{A-CO_2} \ln(y_A^{A-CO_2}) \right]$$

Capturing CO₂ From Air is Very Difficult, but technically feasible

- Energy scales with dilution!
- Density changes with purity
 $95\% \text{CO}_2 + 5\% \text{N}_2 = 681 \text{ kg/m}^3$
 $80\% \text{CO}_2 + 20\% \text{N}_2 = 343 \text{ kg/m}^3$
- $\sim 0.5 \text{ kJ/mol CO}_2$ additional compression energy!



Message #2

**Industrial Emissions May be Low-Hanging Fruit
and Scale with Utilization**

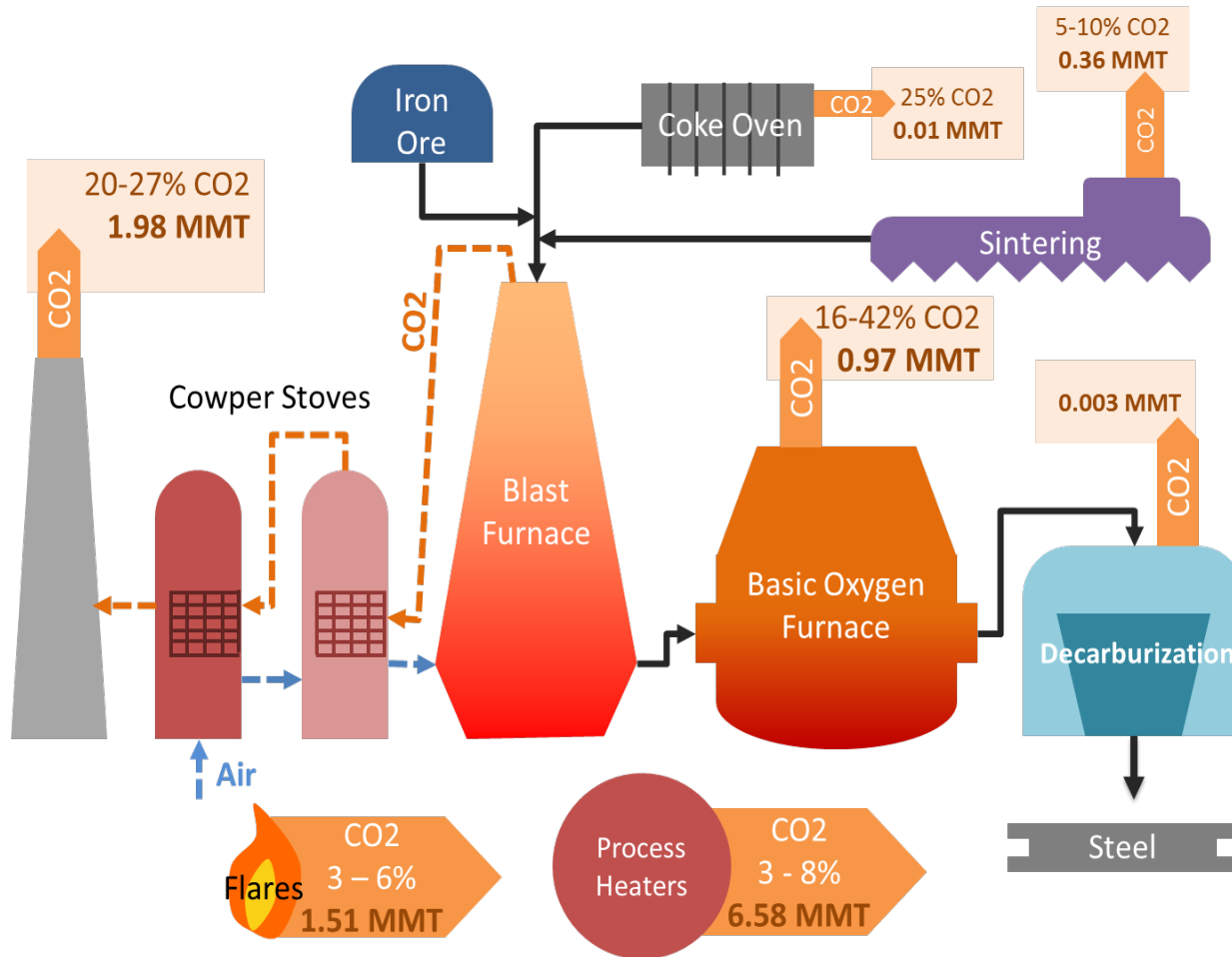
Sherwood Estimates for Separation From Industry

TABLE 2. Minimum work and Sherwood-derived capture cost estimations for various industries.

Source	CO ₂ Content (mol %) ^a	Ref.	Min. Work (kJ/mol CO ₂ Captured) ^b	Cost (US\$/t CO ₂ Captured)	Literature Estimates (2016 US\$/t CO ₂ Captured)	Ref.
Aluminum	4 – 10	6, 12, 13	8.2 – 10.8	45.8 – 65.6	68.2 ^{c,d} – 76.3 ^{c,e}	14
Ammonia	30 – 99.9+	15	0.0 – 5.0	0.0 ^f – 29.0	21.2 ^{c,g}	16
Carbonates	20	17	6.2	36.0		
Cement	14 – 33	18, 19	4.7 – 7.3	28.1 – 39.2	57.3 ^{c,g} , 68.4 ^{c,g} , 54.8 ^h – 95.3 ⁱ ,	16, 20, 21
Ethanol	99.9+	22	0.0	0.0 ^f		
Ferroalloys	8 – 10	23-25	8.3 – 8.9	46.3 – 50.6		
Glass	7 – 12	23, 26	7.7 – 9.3	44.4 – 54.9		
Iron and Steel	20 – 27	18, 27	5.5 – 6.2	31.4 – 34.2	21.7 ^j – 24.4 ^j , 32.6 ^k – 44.0 ^k , 57.3 ^{c,g}	16, 28, 29
Lead	15	30	7.1	40.5		
Lime	20	23	6.2	34.4		
Magnesium	15	23, 31	7.1	40.7		
Petrochemicals	30 – 99.9+	27, 32	0.0 – 5.0	0.0 ^f – 28.6		
Pulp and Paper	8	33	8.9	48.0	31.1 ^l – 35.0 ^l	34
Refining	3 – 20	35, 36	6.2 – 11.7	33.5 – 70.4		
Silicon Carbide	8	37	8.9	51.4		
Soda Ash	36 – 40	38	4.0 – 4.4	25.6 – 26.7		
TiO ₂	13	39	7.5	41.2		
Zinc	15	30, 40	7.1	40.2		
Natural Gas	3 – 5	9, 32	10.3 – 11.7	57.2 – 69.9		
Petroleum	3 – 8	9, 18	8.9 – 11.7	47.0 – 69.0		
Coal	10 – 15	32, 41, 42	7.1 – 8.3	36.5 – 42.7		

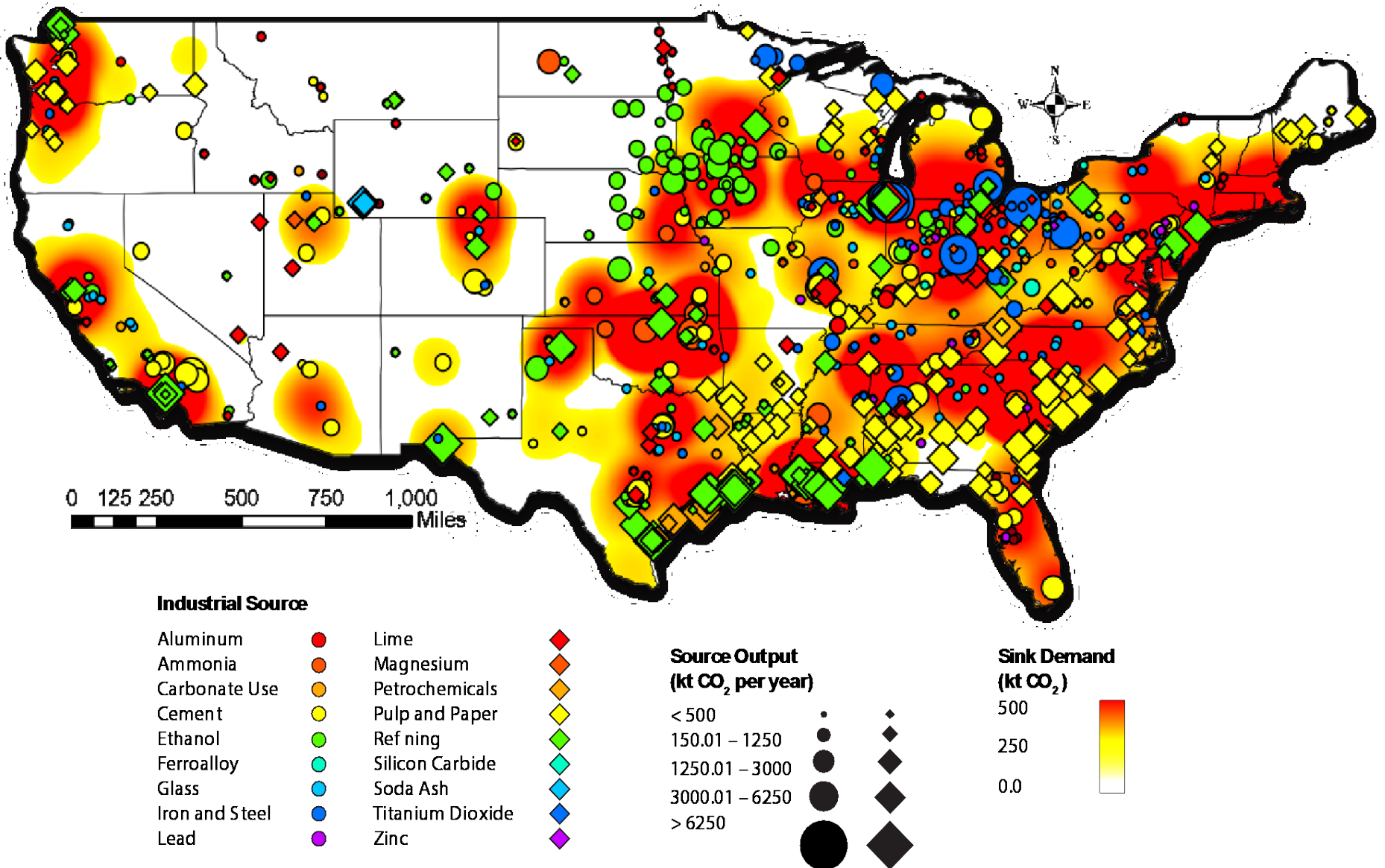
^a Range in composition due to different processes or different capture points within the same process. When not directly reported, values were estimated from a complete mass balance assuming NG fuel and 15% excess air; ^b calculated assuming 99.5% purity and 90% capture, Ref. ⁹; ^c includes cost of compression; ^d calculated at 10% CO₂ purity; ^e calculated at 4% CO₂ purity; ^f for near pure streams, separation costs are considered in the compression and dehydration stage; ^g includes costs for transport and storage; ^h oxycombustion; ⁱ post-combustion MEA; ^j selexol capture from the blast furnace; ^k post-combustion capture from blast furnace; ^l capture using pre-combustion shift technology

Rethinking Industry-Based Emissions

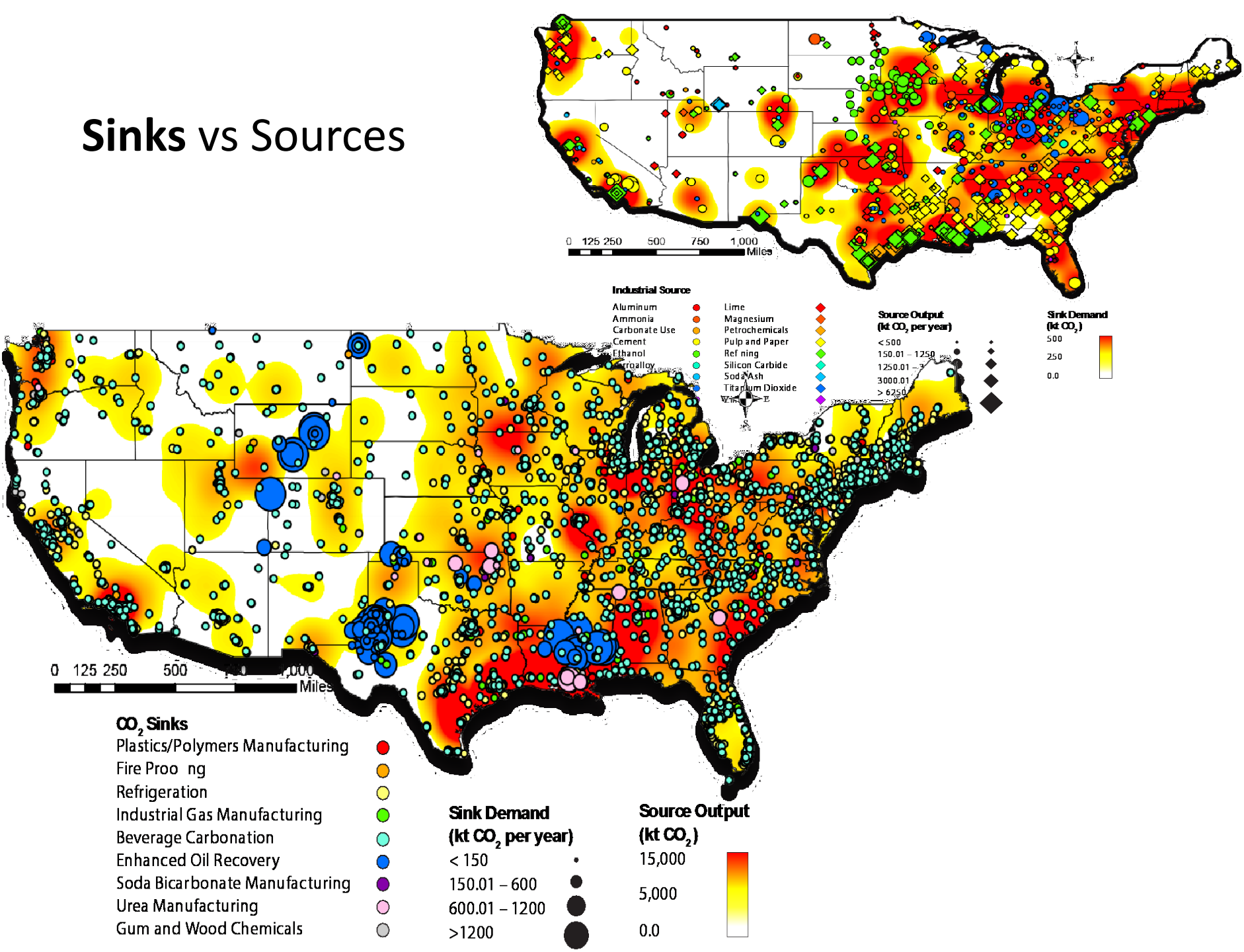


Example – US Steel Corporation plant in Gary, Indiana

Sources vs Sinks



Sinks vs Sources



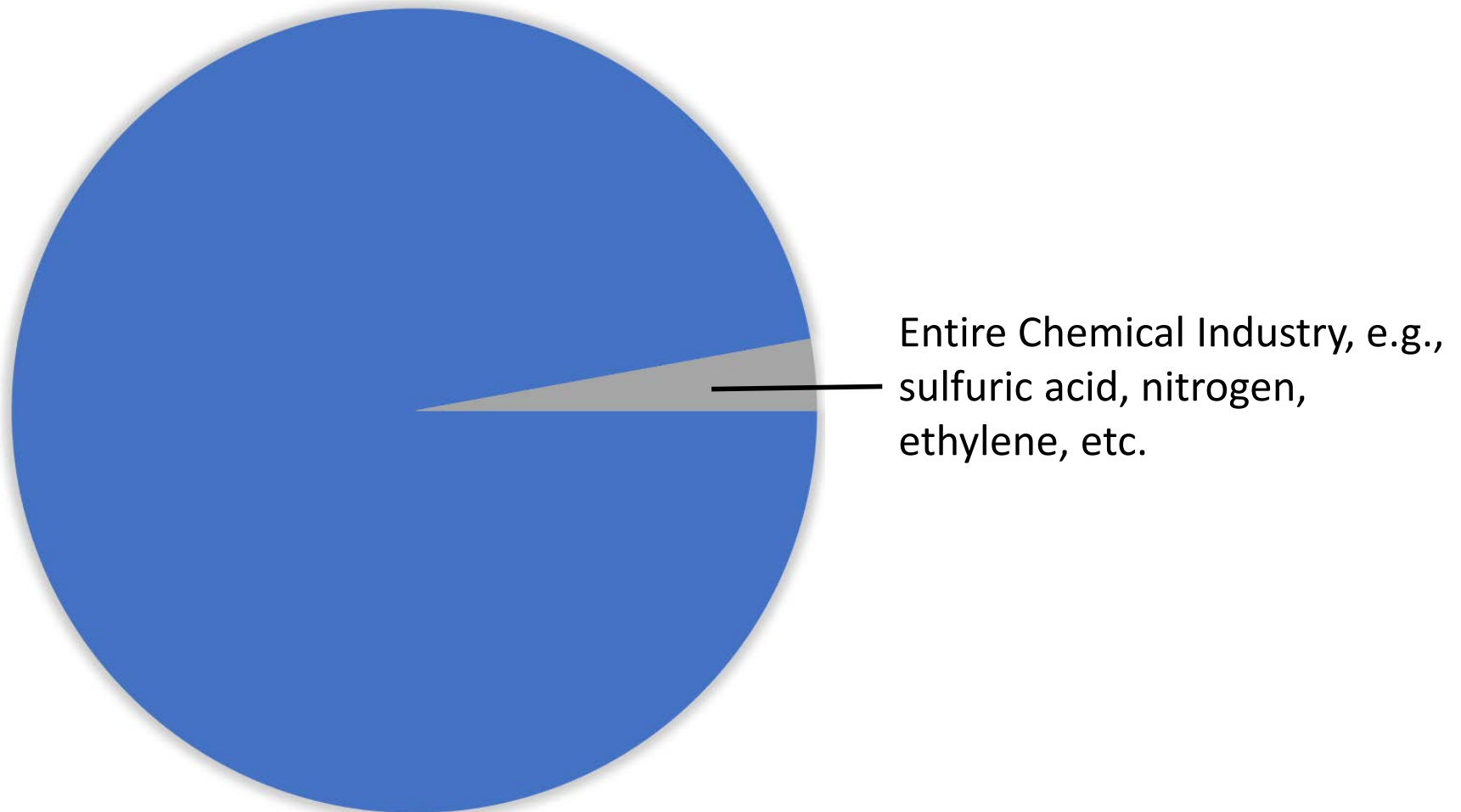
Message #3

**Using CO₂ as a Chemical Feedstock May have a
Minor Impact on Emissions**

...but Several Opportunities Exist on a Gt-Scale



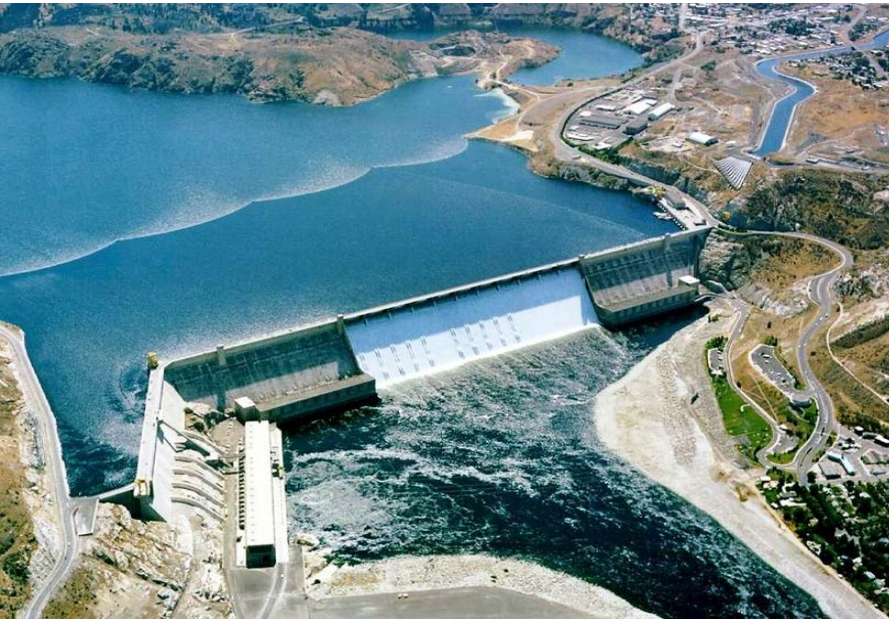
Chemical Industry Represents 3% of Emissions Globally



Products that Have an Impact on a Gt-Scale

- **Steel** market in 2018 is predicted to be ~ **1.55 Gt**
- **Construction aggregates** ~ **53 Gt in 2017**, up from 37 Gt in 2010, in terms of concrete, this represents building a sidewalk around the equator 5,000 times! or ~ 1,400 Three Gorges dams!

Grand Coulee Dam 21.3 Mt concrete



Three Gorges Dam 36.2 Mt concrete



Note: Concrete = cement + water + sand and gravel (aggregate) and cement industry ~ 4 Gt (2015) – rule of thumb is that 6-7x aggregate required; **Reference:** USGS

Message #4

**First and Foremost, Mitigation Efforts Must
Increase by at least 100x**

CCS Demonstration Projects

There are ~ 100 projects globally that are operating, have operated, or are under construction

- Scale: ~1 Mt CO₂/yr (compared to ~ 35 Gt CO₂/yr emitted)
- CO₂ Sink: Geologic formations coupled with enhanced oil recovery
- EOR CO₂ storage to date – 560 Mt CO₂
- ARI Report – EOR may provide a CO₂ market of up to **7.5 GtCO₂** bet/ 2016-2030 (value of ~ \$260B)

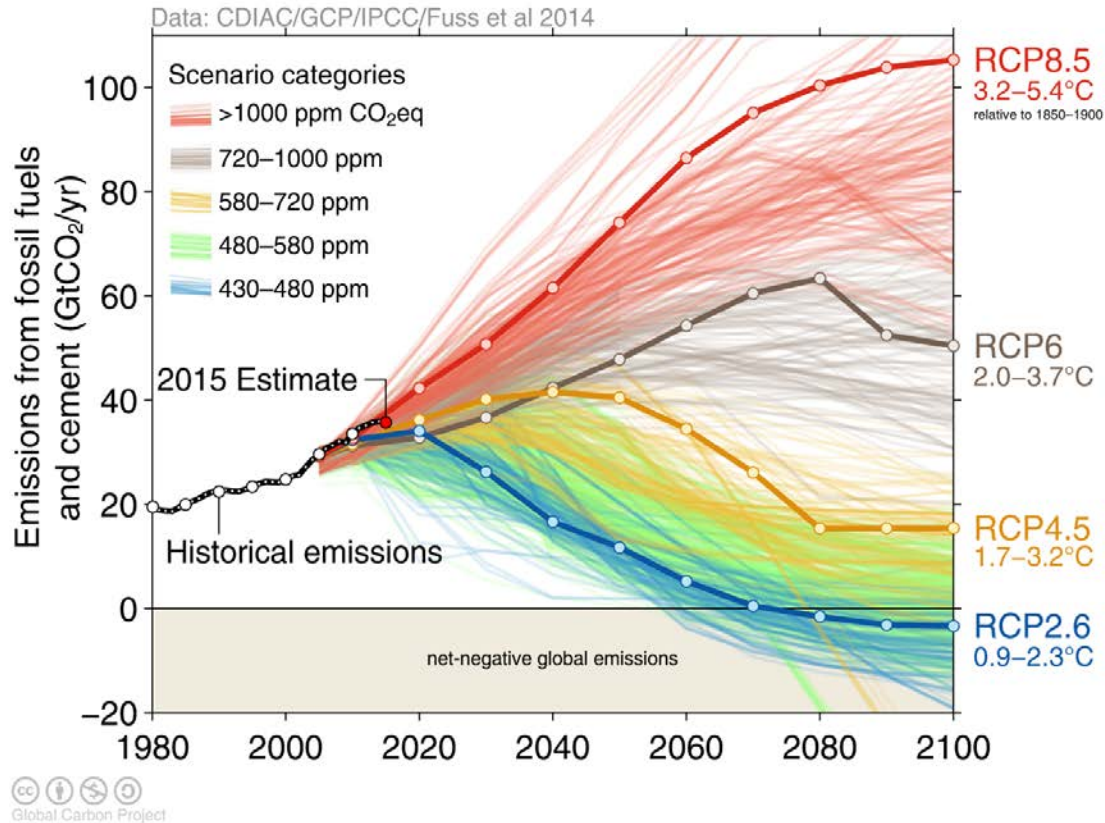




Message #5

**Negative Emissions will be Required
to Prevent 2 °C warming by 2100**

Climate Models Include NETs



Parting Message

We need everything and we need it now
e.g., fuel switching, CCS, renewables, negative emissions, etc.