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Using different flavours of oxygen to measure biological production from ship-based and autonomous platforms

Jan Kaiser

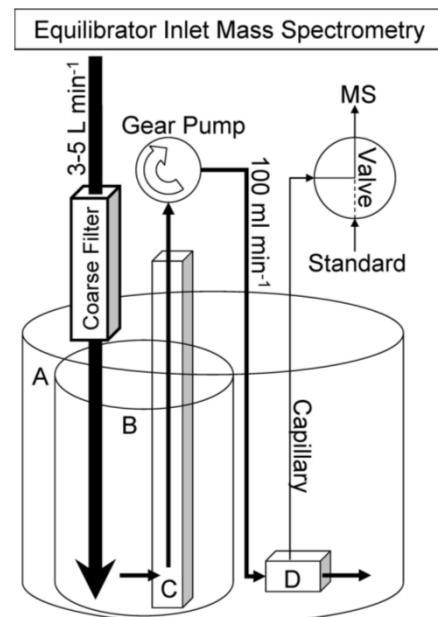
University of East Anglia
Centre for Ocean and Atmospheric Sciences
School of Environmental Sciences
Norwich
United Kingdom

Flavours of oxygen

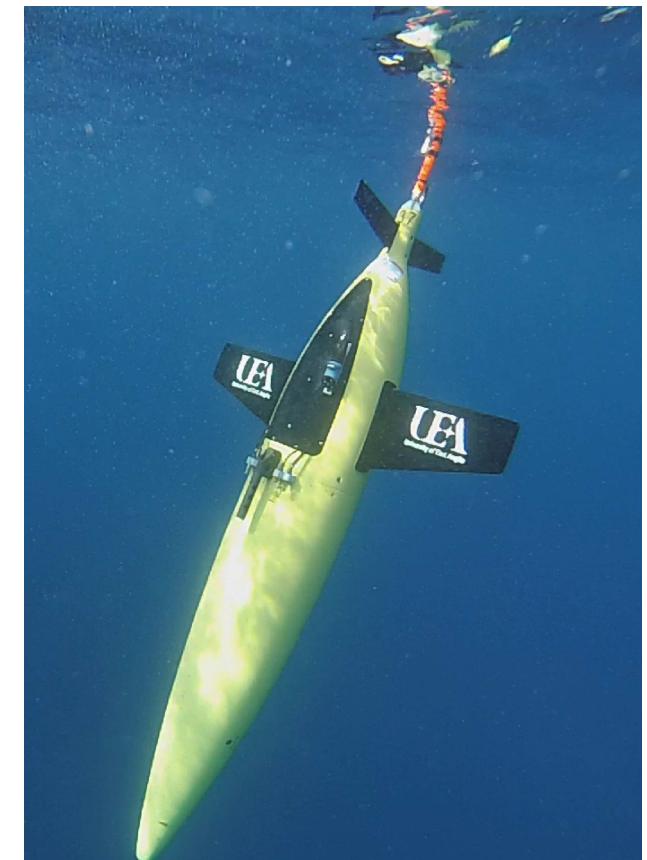
- molecular dioxygen: O_2
- oxygen triple isotopologues: $^{16}O_2$, $^{16}O^{18}O$, $^{16}O^{17}O$
- abiotic gas exchange analogue: O_2 and Ar
- "abiotic" deep-water mixing analogue: O_2 and N_2O

Sampling and analysis

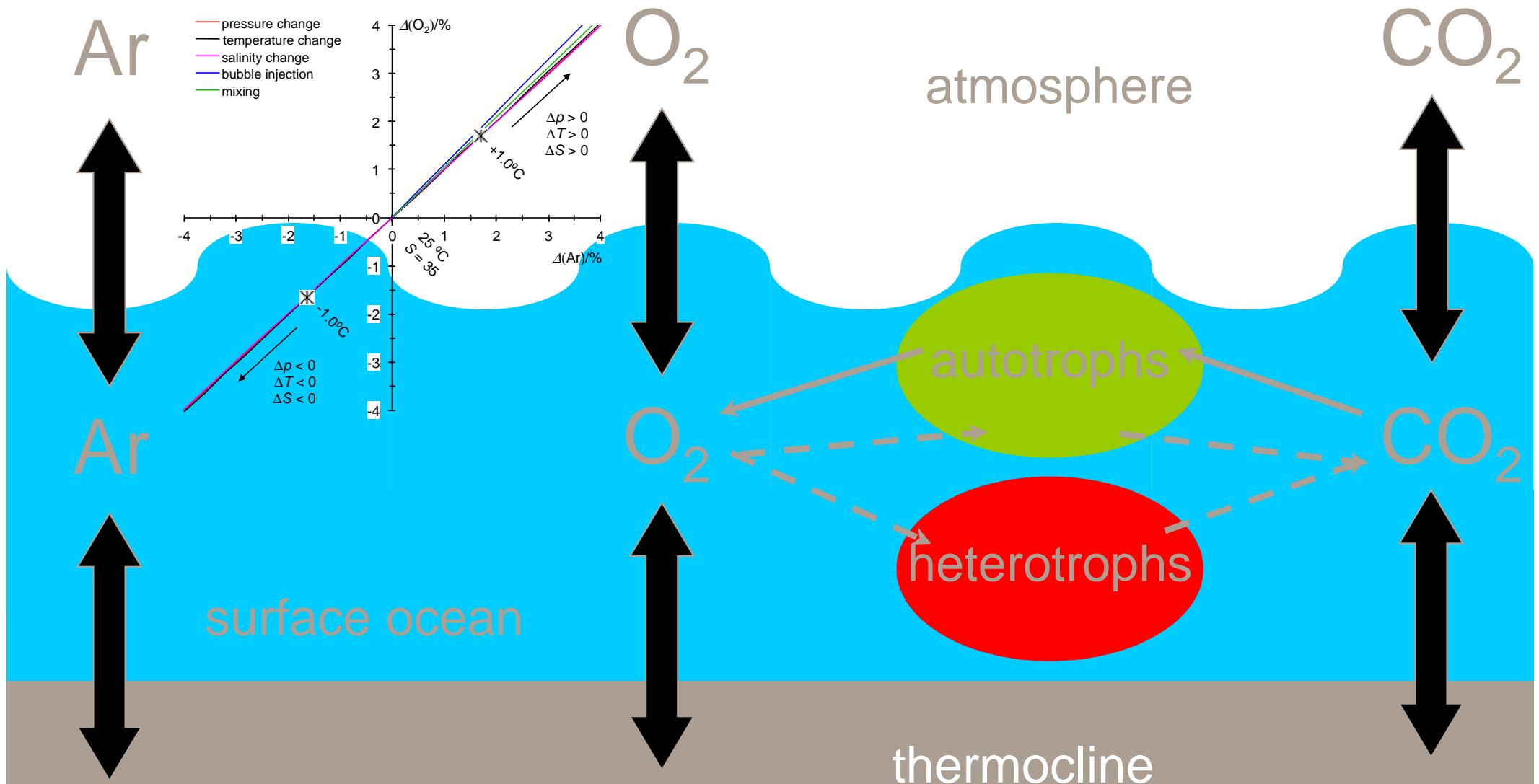
- continuous O₂ concentration measurements by **optode**
- continuous O₂/Ar ratio measurements by **MIMS** or **EIMS**
- O₂ isotopologues by **isotope-ratio mass spectrometry (IRMS)**
- N₂O concentrations by **laser cavity absorption spectroscopy**



Platforms

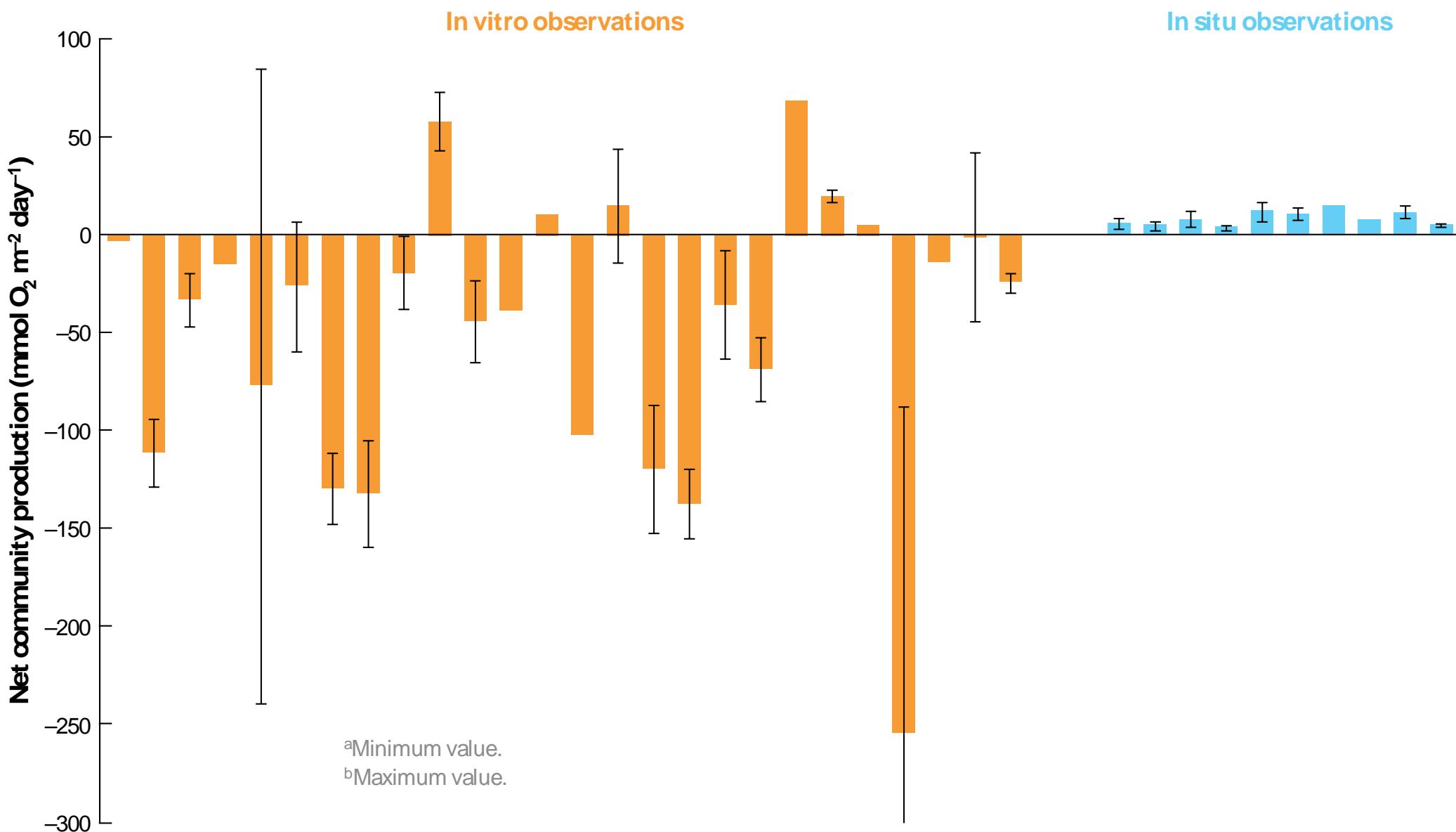


Production (P), respiration (R) [$N = P - R$], gas exchange, mixing



In vitro vs. in situ net production

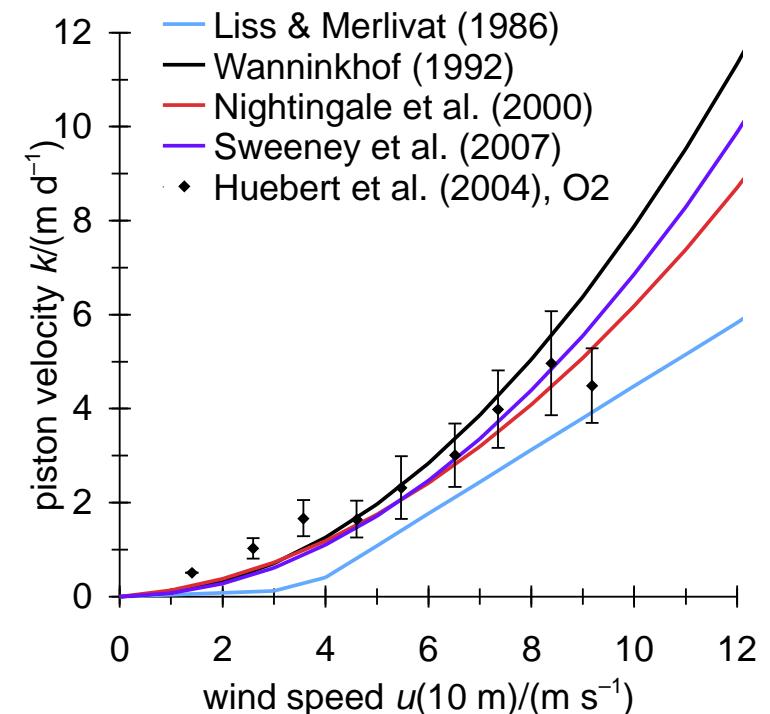
Williams et al. Ann. Rev. Mar. Sci. 2013



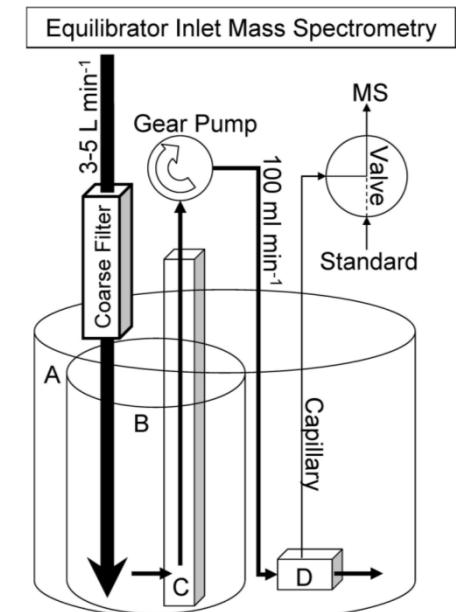
Advantages & disadvantages

- + in principle, unambiguous measurement of N and P
- + no sampling biases (bottle effects, temperature, light)
- + ease of sampling (non-research vessels, no incubations)
- + integrates over O_2 residence time in the mixed layer (days to weeks)
- + scalable / high spatial coverage possible

- uncertainty of gas exchange coefficient ($\pm 20\%$)
- currency is O_2 , not C
 - photosynthetic coefficient?
 - P includes other non-carbon fixing, H_2O -splitting reactions
- integrates vertically over mixed layer depth, but may include contributions from below



Using Ar to correct for the physical O₂ component



Net community production $N = P - R$

O₂ mass balance:

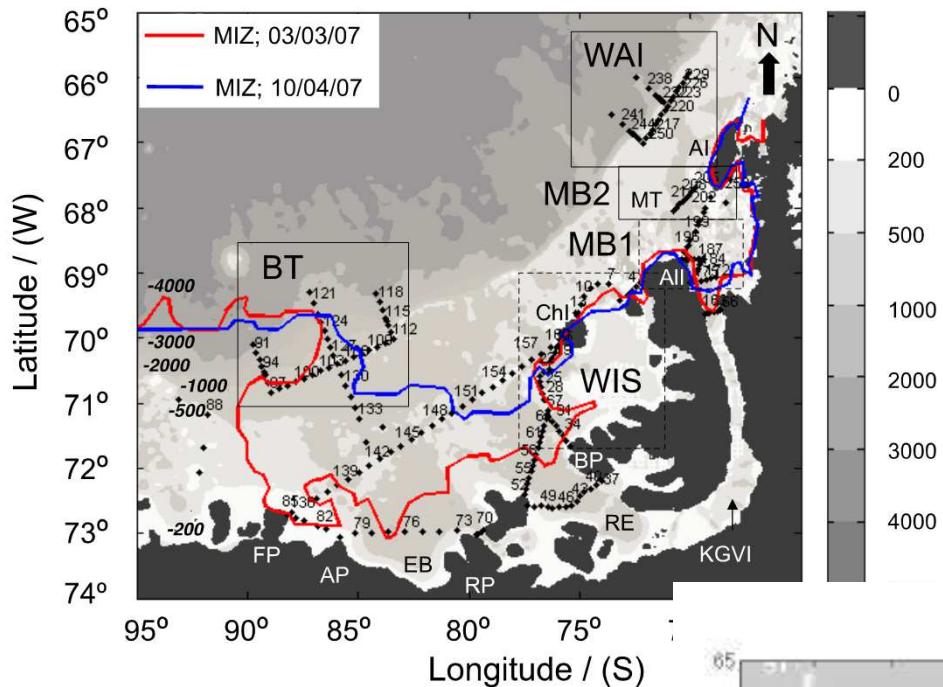
$$z_{\text{mix}} \frac{\partial c(O_2)}{\partial t} = N - k(O_2) c_{\text{sat}}(O_2) \Delta(O_2) + F_{\text{inj}} \chi(O_2) + F_{\text{exch}} \frac{\alpha(O_2) \chi(O_2)}{Sc(O_2)} + K_z \left. \frac{\partial c(O_2)}{\partial z} \right|_{z_{\text{mix}}}$$

Biological O₂ flux:

$$F_{\text{bio}}(O_2/\text{Ar}) = k c_{\text{sat}}(O_2) \Delta(O_2/\text{Ar}) = N + K_z \left. \frac{\partial c_{\text{sat}}(O_2) \Delta(O_2/\text{Ar})}{\partial z} \right|_{z_{\text{mix}}}$$

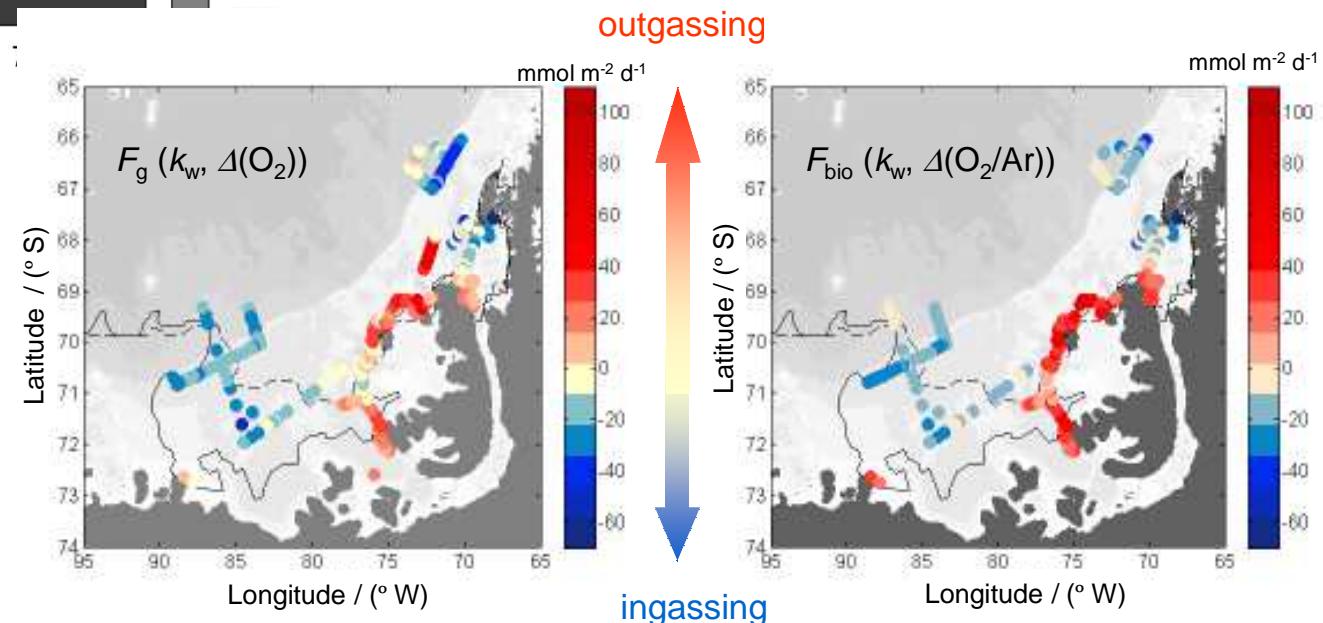
$$N \approx F_{\text{bio}}(O_2/\text{Ar})$$

Production in the Bellingshausen Sea



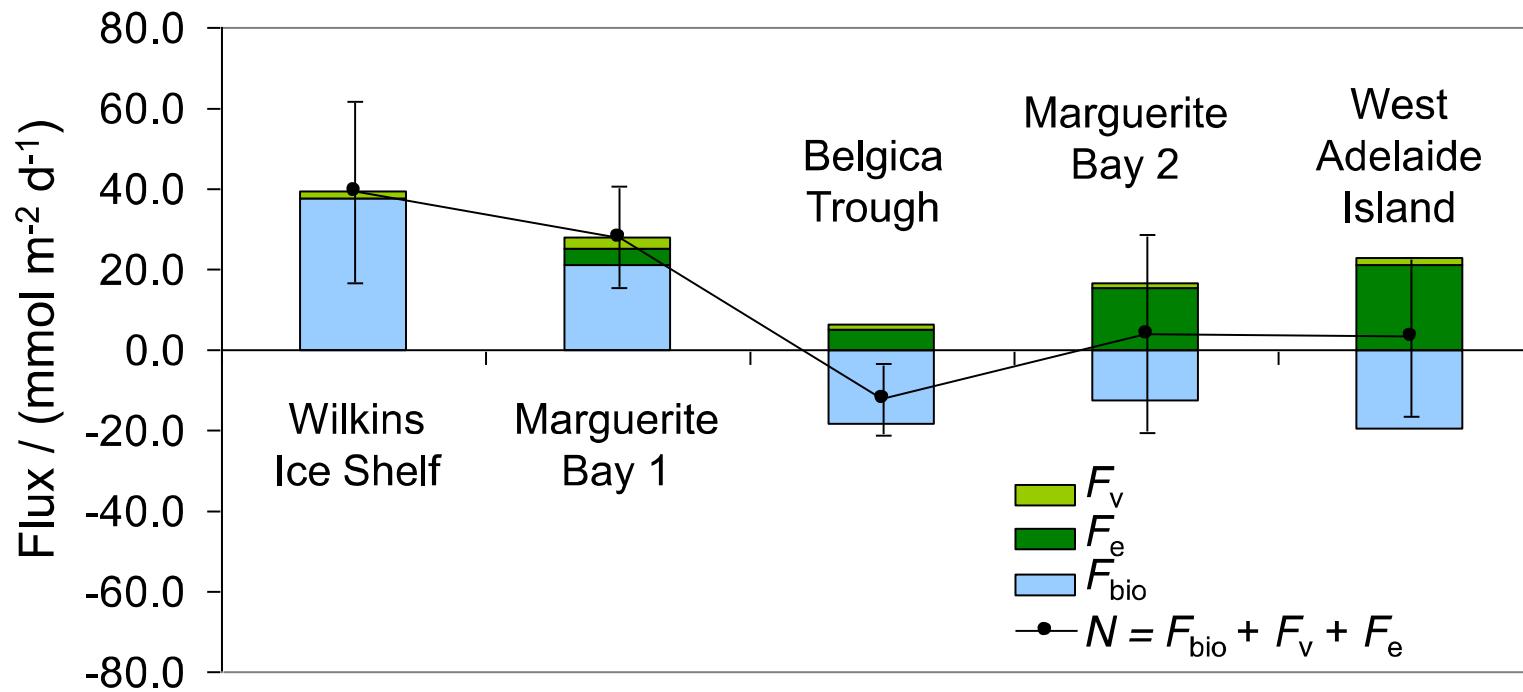
Mixture of open
ocean and coastal
sampling locations

Effect of Ar correction:

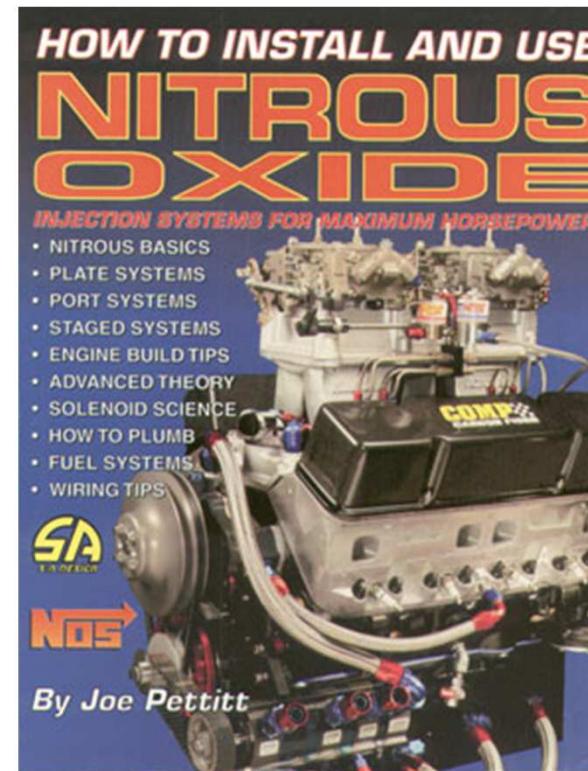


Entrainment correction "removes" apparent net heterotrophic areas

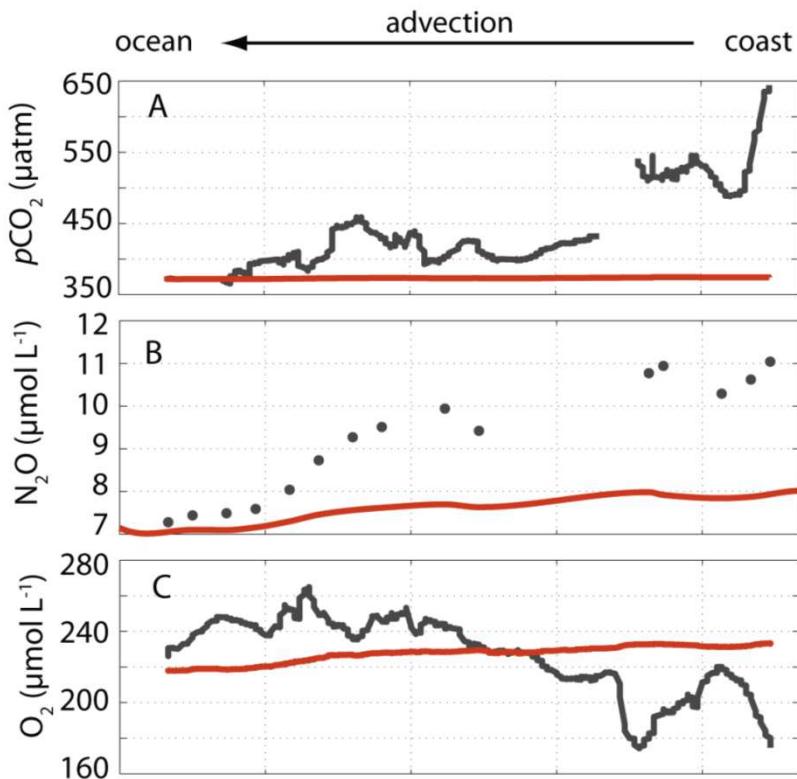
$$F_e = z_{miz} \quad \frac{\Delta c}{\Delta t} = \frac{1}{2} \frac{(\Delta z_{mix})^2}{\Delta t} \left. \frac{\partial c(O_2)}{\partial z} \right|_{oxy} .$$



Using N₂O to correct for vertical fluxes



Using N₂O as "abiotic" analogue to correct for vertical O₂ fluxes



Upwelling brings up

- high N₂O
- high CO₂
- low O₂
- high nutrient

Re-equilibration by gas-exchange and net production.

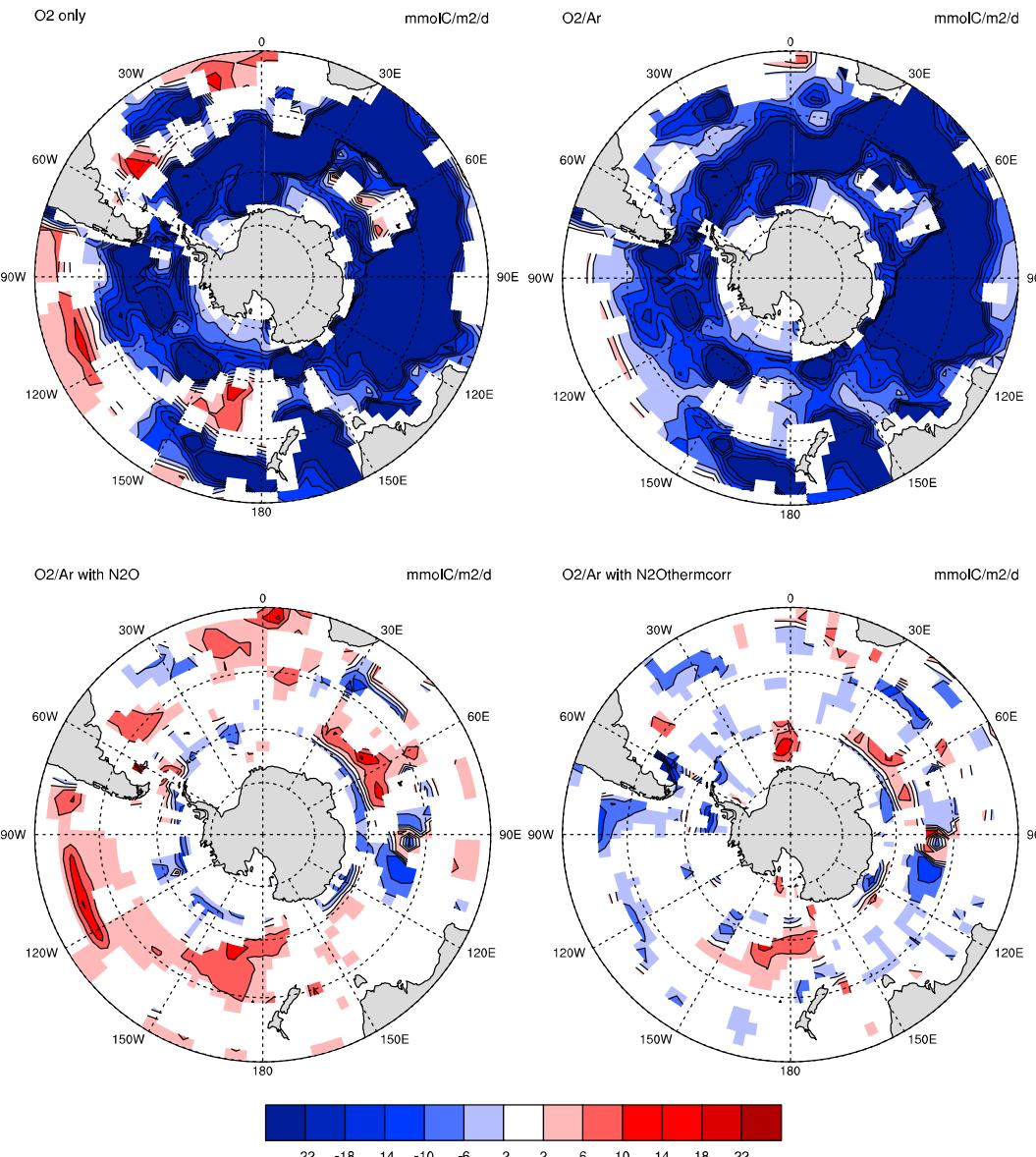
N₂O has negligible surface sources and sinks:

$$z_{\text{mix}} \frac{dc(\text{N}_2\text{O})}{dt} = -k(\text{N}_2\text{O})[c(\text{N}_2\text{O}) - c_{\text{equ}}(\text{N}_2\text{O})]$$

$$z_{\text{mix}} \frac{dc(\text{O}_2)}{dt} = N(\text{O}_2) - k(\text{O}_2)[c(\text{O}_2) - c_{\text{equ}}(\text{O}_2)]$$

$$N(\text{O}_2) = k(\text{N}_2\text{O})[c(\text{N}_2\text{O}) - c_{\text{equ}}(\text{N}_2\text{O})] \frac{dc(\text{O}_2)}{dc(\text{N}_2\text{O})} + k(\text{O}_2)[c(\text{O}_2) - c_{\text{equ}}(\text{O}_2)]$$

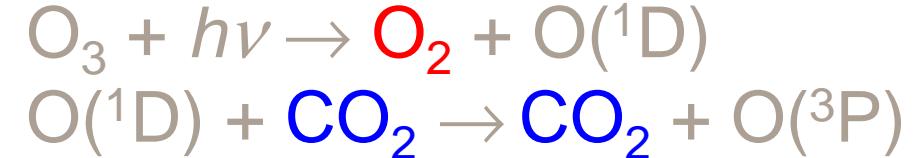
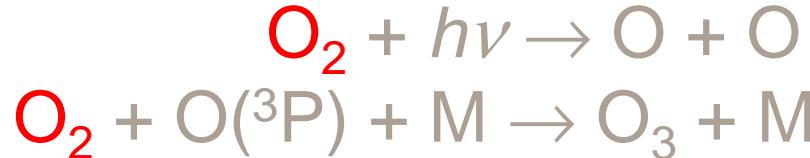
Correction of $F_{\text{bio}}(\text{O}_2/\text{Ar})$ for vertical mixing using N_2O to give $N(\text{O}_2/\text{Ar})$



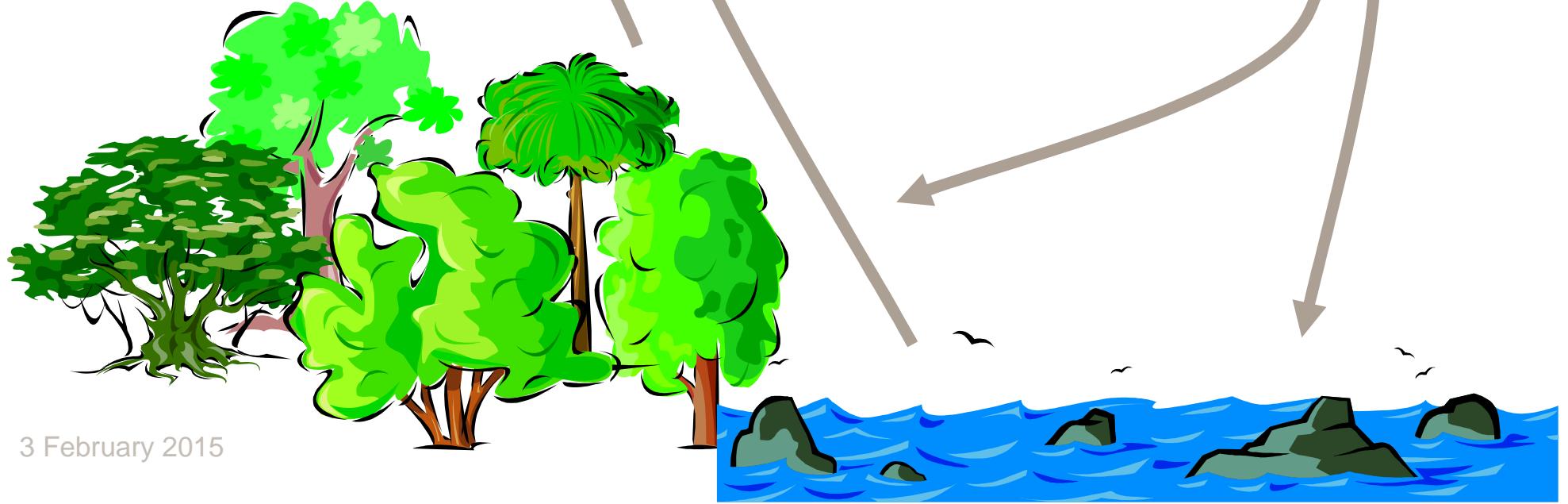
Gross production using oxygen triple isotopologues ($^{16}\text{O}_2$, $^{16}\text{O}^{17}\text{O}$, $^{16}\text{O}^{18}\text{O}$)



Oxygen isotope transfer from O_2 to CO_2 via O_3



tropopause



^{17}O excess, $\Delta(^{17}\text{O})$, of photosynthetic O_2

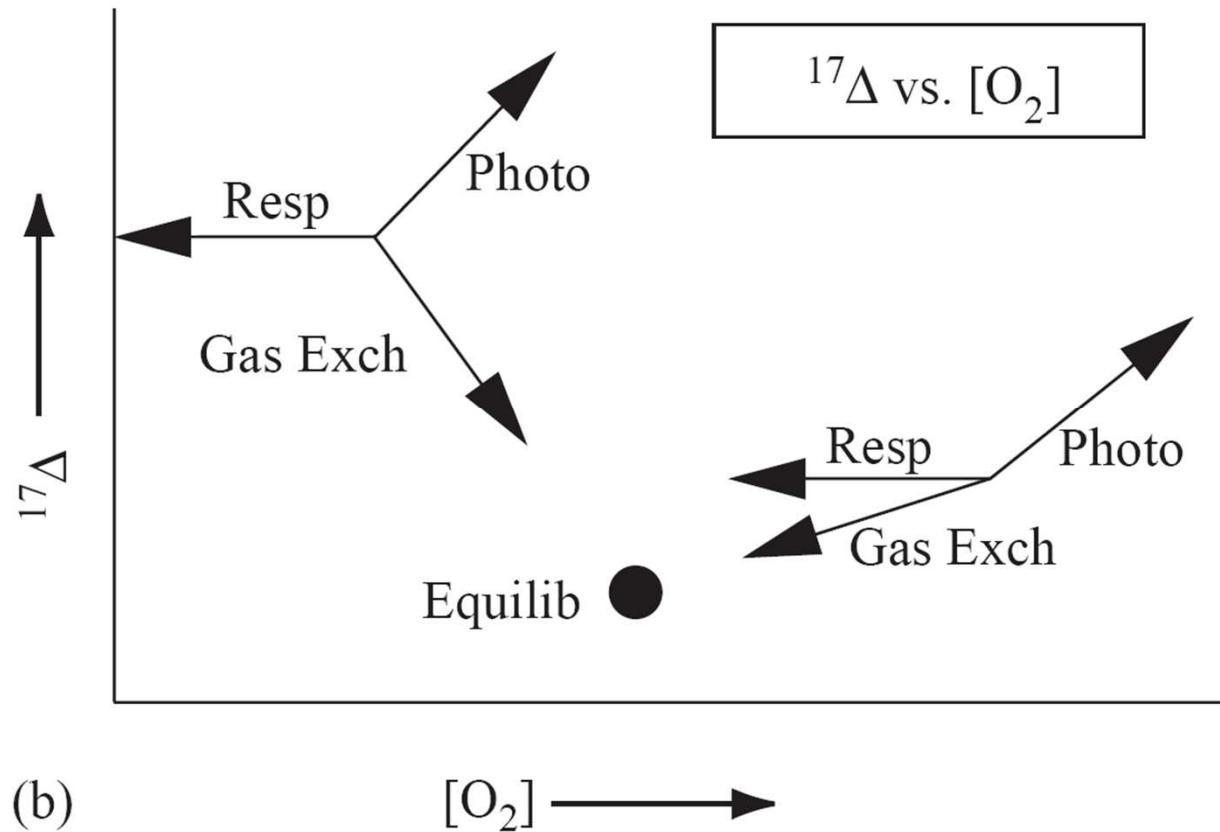
$$\Delta(^{17}\text{O}) = \delta(^{17}\text{O}) - \lambda\delta(^{18}\text{O})$$

Choose $\lambda = \gamma_R = 0.5179 = {}^{17}\varepsilon_R / {}^{18}\varepsilon_R$,
where ε_R is the respiratory kinetic isotope fractionation

Reference: tropospheric Air- O_2

photosynthetic O_2 : $\Delta_{\max}(^{17}\text{O}) = 249 \text{ ppm (???)}$
 $(180 \text{ to } 264 \text{ ppm})$

Effect of photosynthesis, respiration and gas exchange on O_2 analogues



Calculating the ratio of gross O₂ gross production P to gross O₂ influx: $g = P / (kc_{\text{sat}})$

Luz & Barkan
(2000):

$$g = \frac{P}{kc_{\text{sat}}} \approx \frac{\Delta(^{17}\text{O}) - \Delta_{\text{sat}}(^{17}\text{O})}{\Delta_{\text{max}}(^{17}\text{O}) - \Delta(^{17}\text{O})}$$

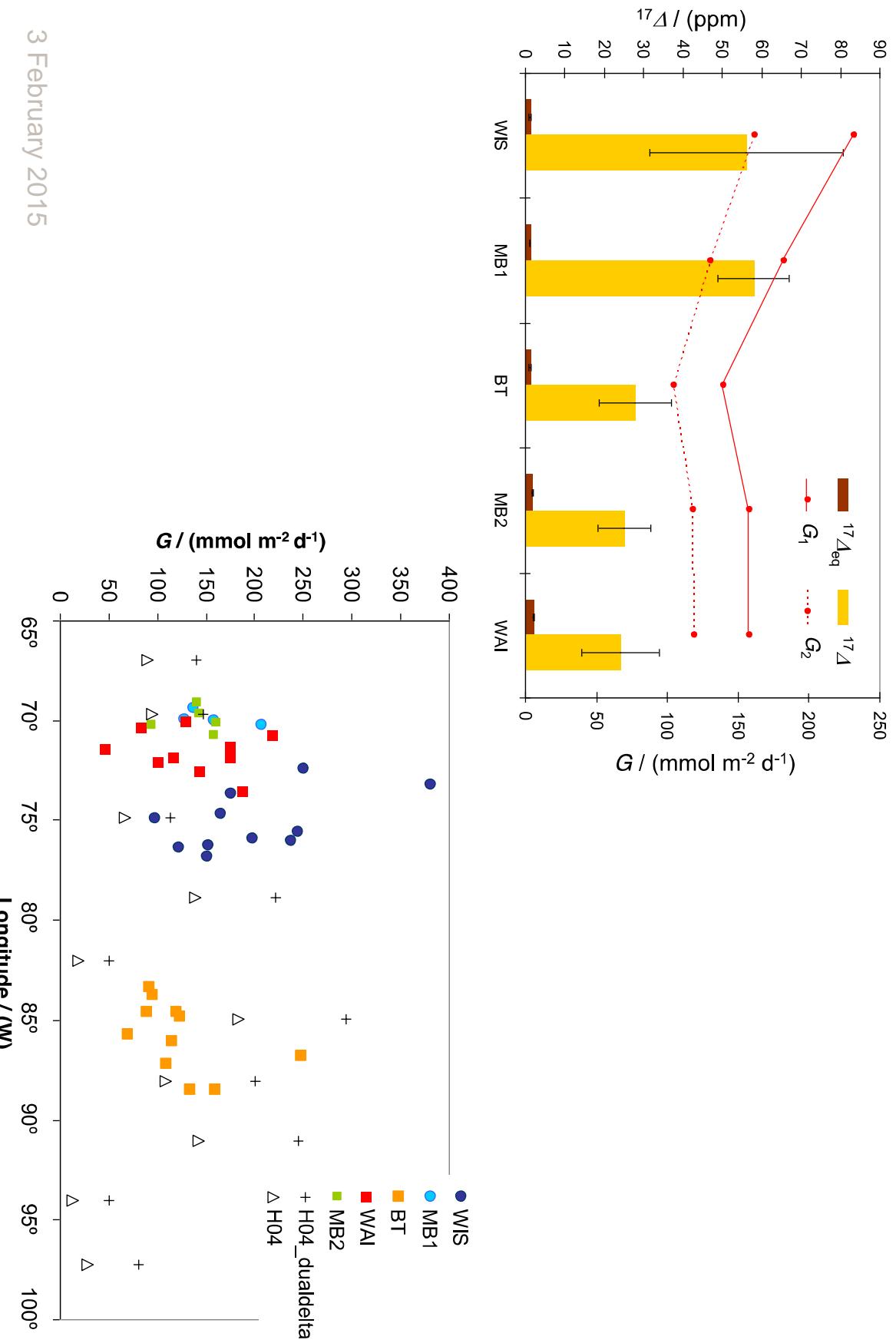
Prokopenko et
al. (2011):

$$g = \frac{\frac{^{17}\delta - ^{17}\delta_{\text{sat}}}{1 + ^{17}\delta} - \gamma_R \frac{^{18}\delta - ^{18}\delta_{\text{sat}}}{1 + ^{18}\delta}}{\frac{^{17}\delta_P - ^{17}\delta}{1 + ^{17}\delta} - \gamma_R \frac{^{18}\delta_P - ^{18}\delta}{1 + ^{18}\delta}}$$

Kaiser (2011):

$$g = \frac{(1 + ^{17}\varepsilon_E) \frac{^{17}\delta - ^{17}\delta_{\text{sat}}}{1 + ^{17}\delta} - \gamma_R (1 + ^{18}\varepsilon_E) \frac{^{18}\delta - ^{18}\delta_{\text{sat}}}{1 + ^{18}\delta} + s(^{17}\varepsilon_E - \gamma_R ^{18}\varepsilon_E)}{\frac{^{17}\delta - ^{17}\delta}{1 + ^{17}\delta} - \gamma_R \frac{^{18}\delta_P - ^{18}\delta}{1 + ^{18}\delta}}$$

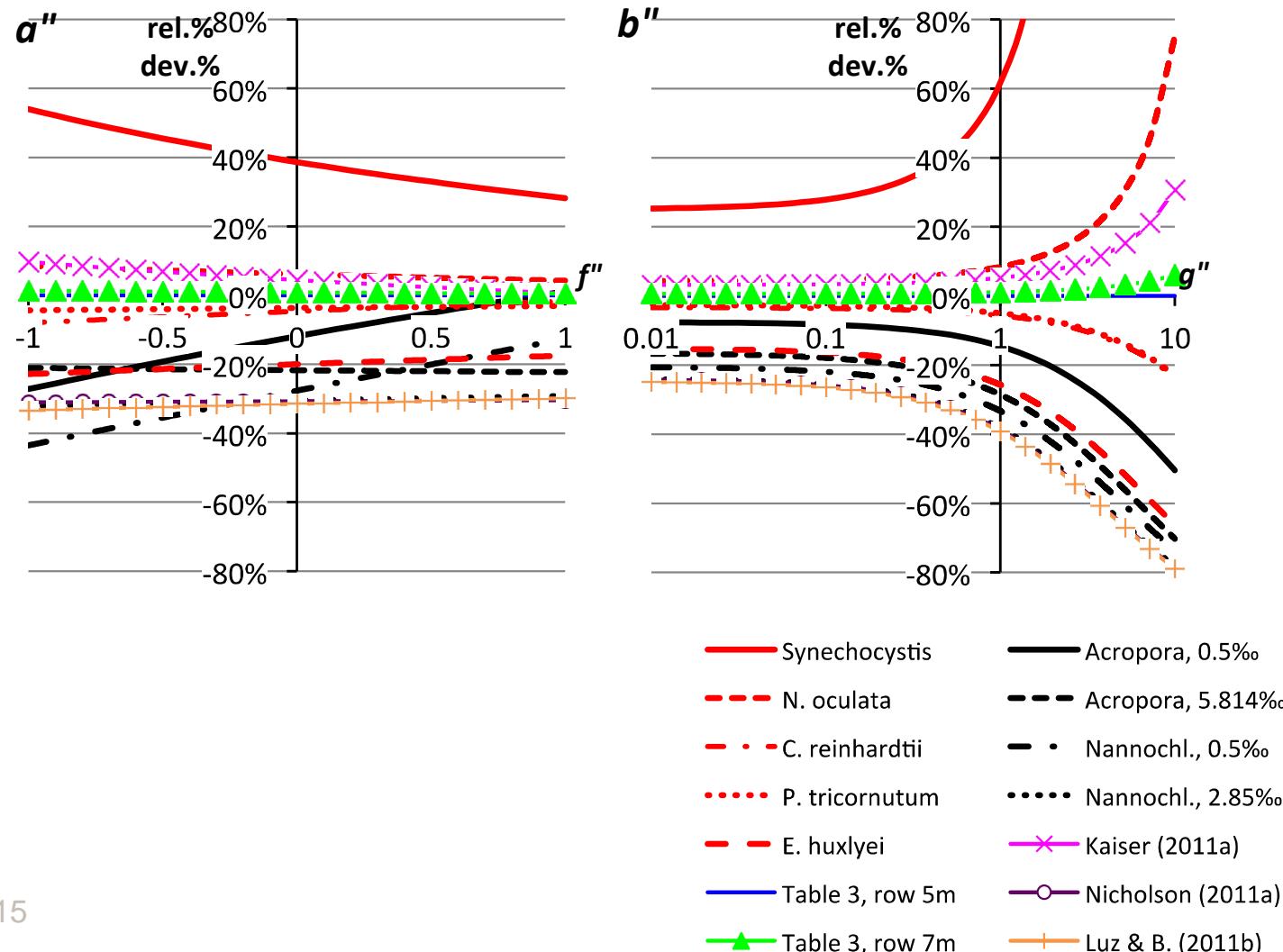
Gross O₂ production in the Bellingshausen Sea



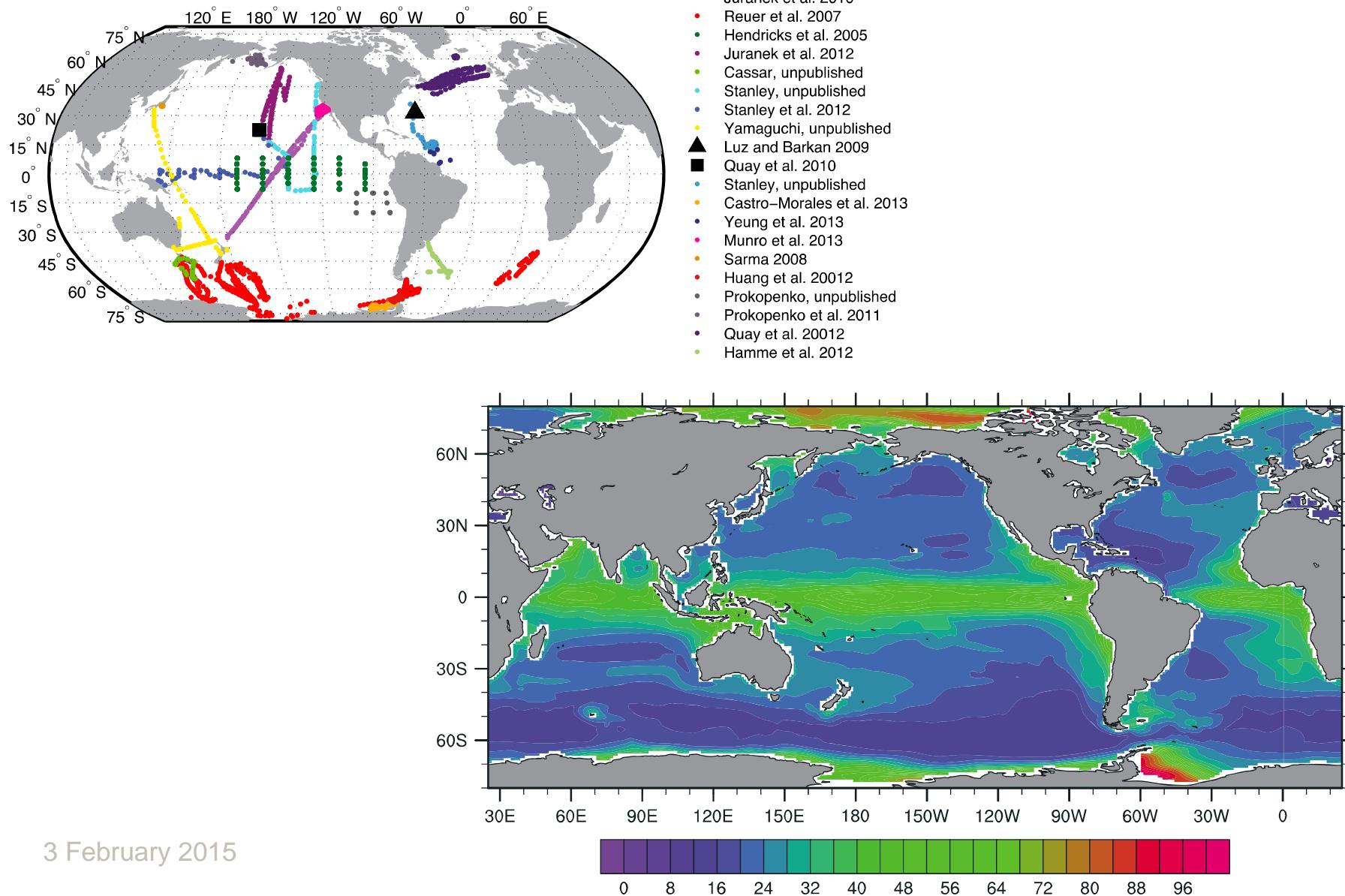
Castro-Morales et al. Biogeosciences 2013;
Hendricks et al. DSR I 2004

Unrecognised systematic uncertainties due to phytoplankton composition

Relative deviation from base case, $\Delta_{\text{max}}(^{17}\text{O}) = 185 \text{ ppm}$



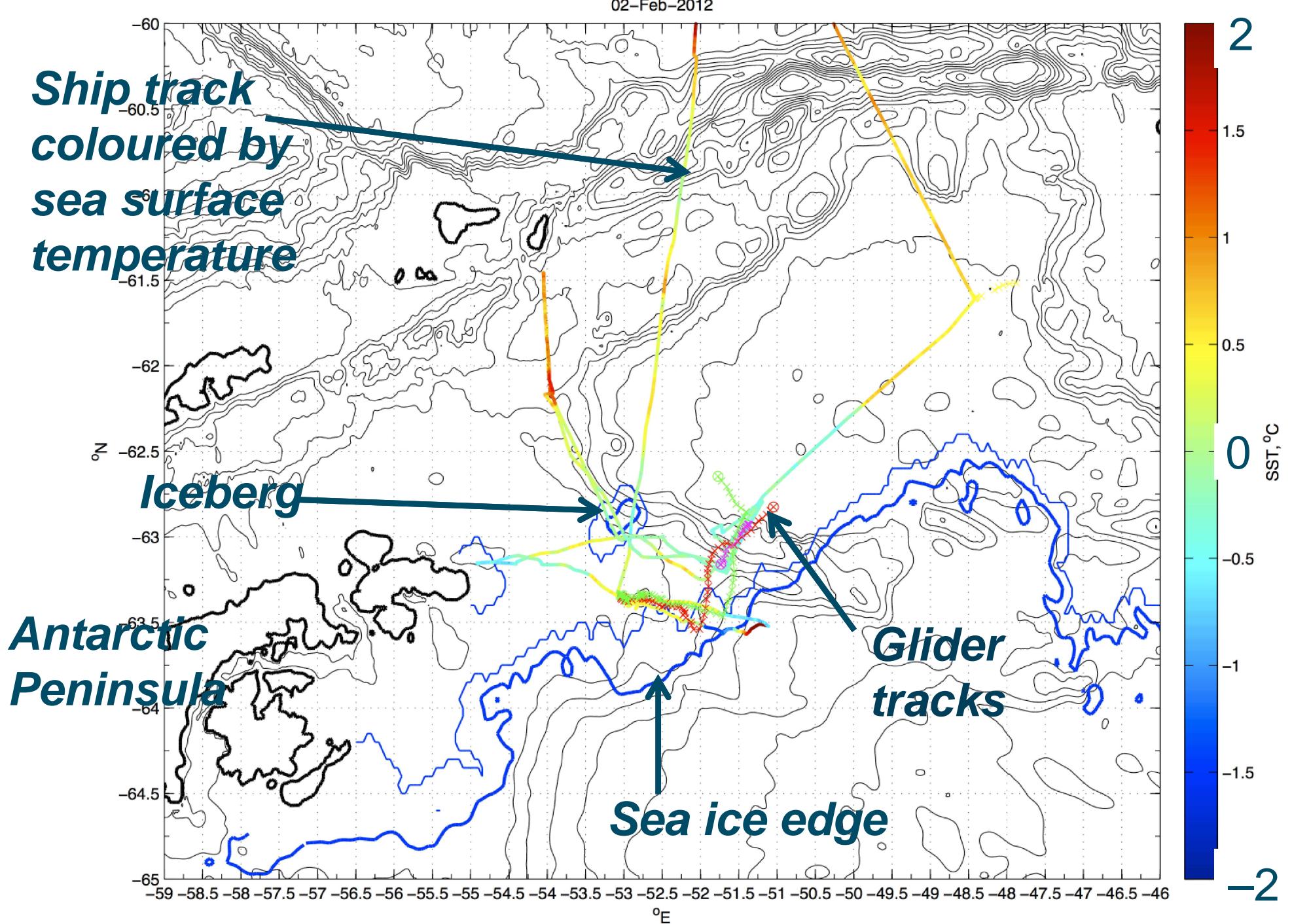
Global measurements and modelling of oxygen triple isotopologues



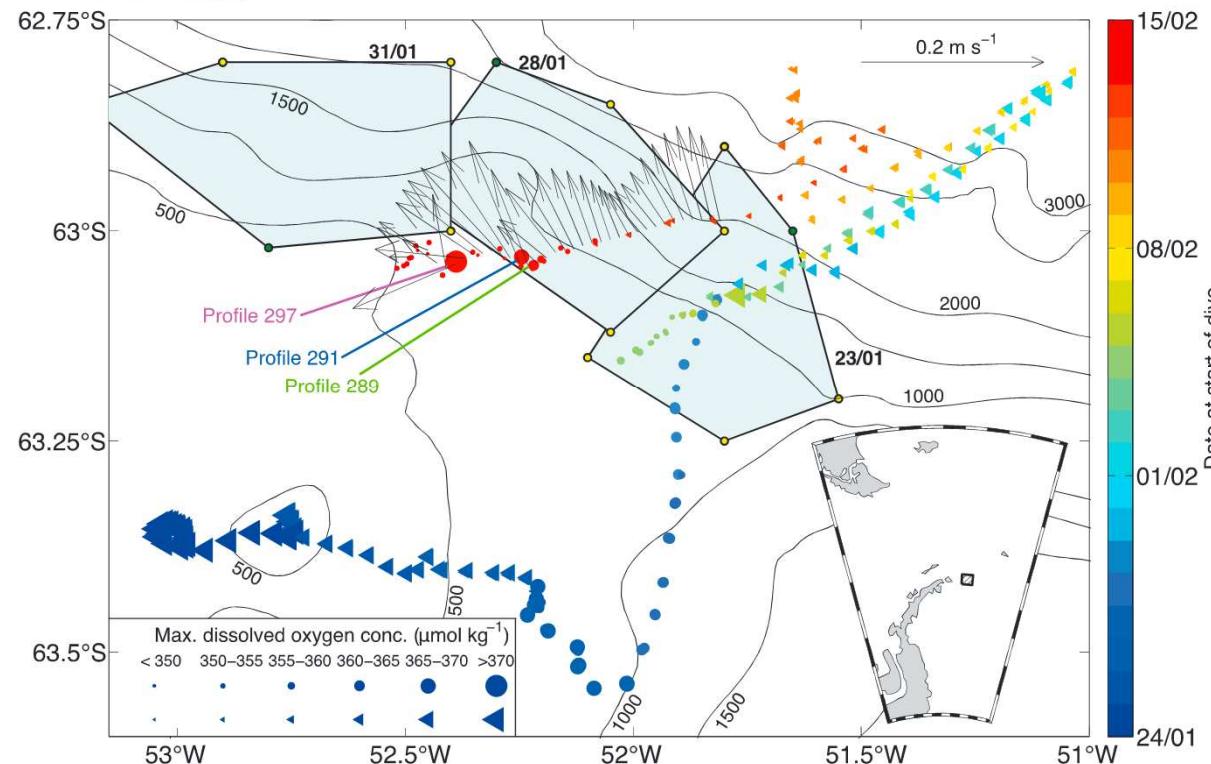
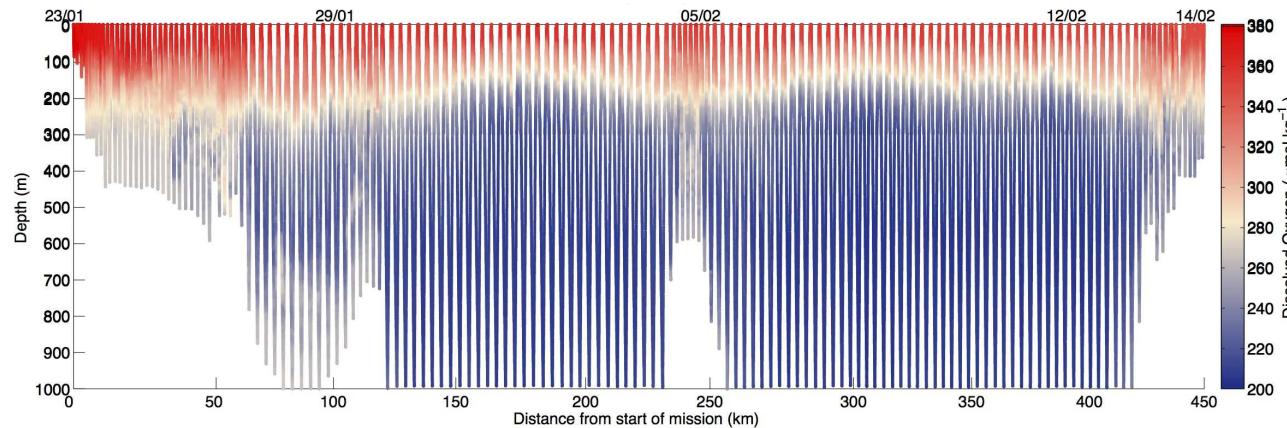
Net community production from underwater ocean gliders



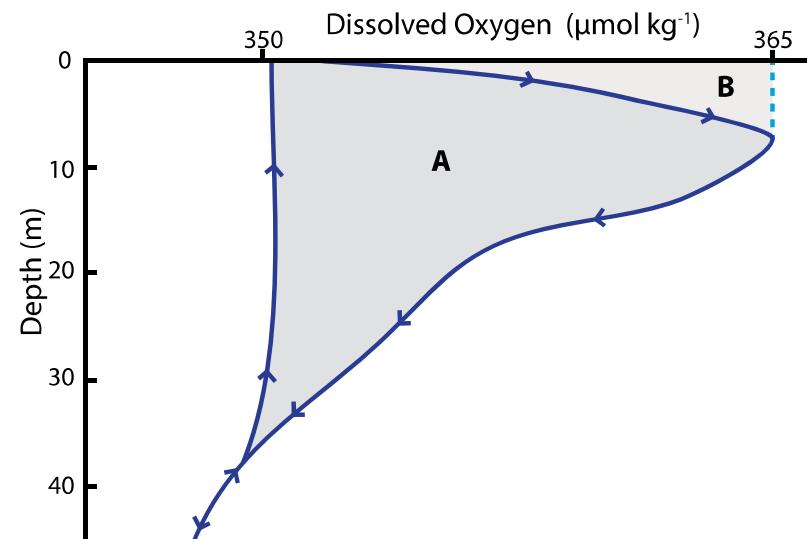
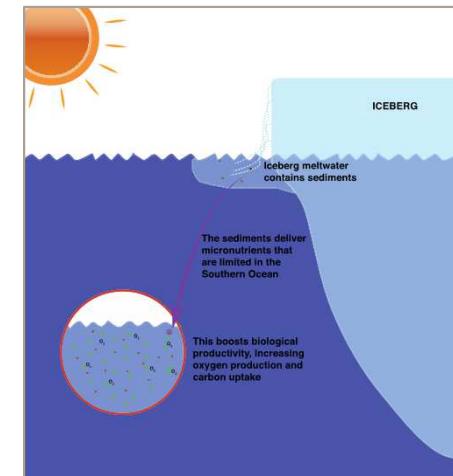
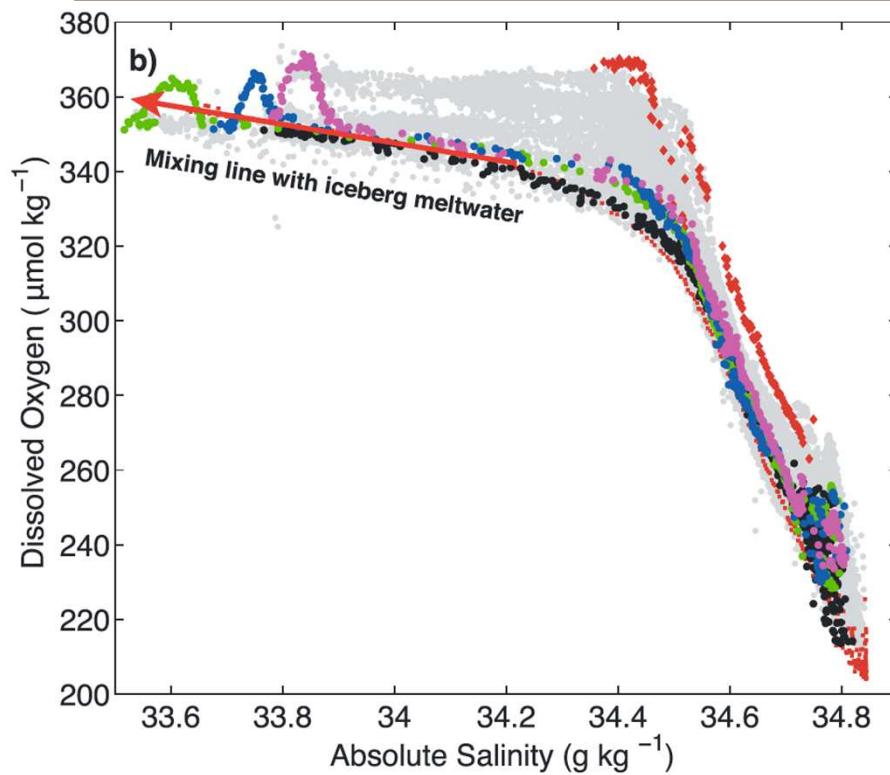
RRS James Clark Ross cruise JR255A "GENTOO", Jan 2012



Glider O₂ measurements



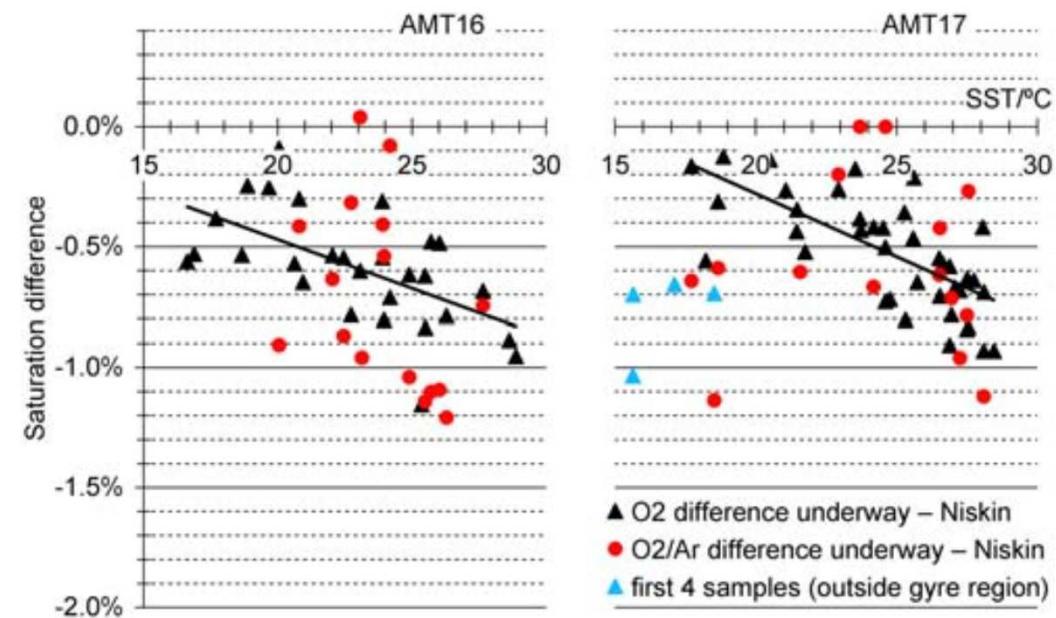
Depth-integrated net community production



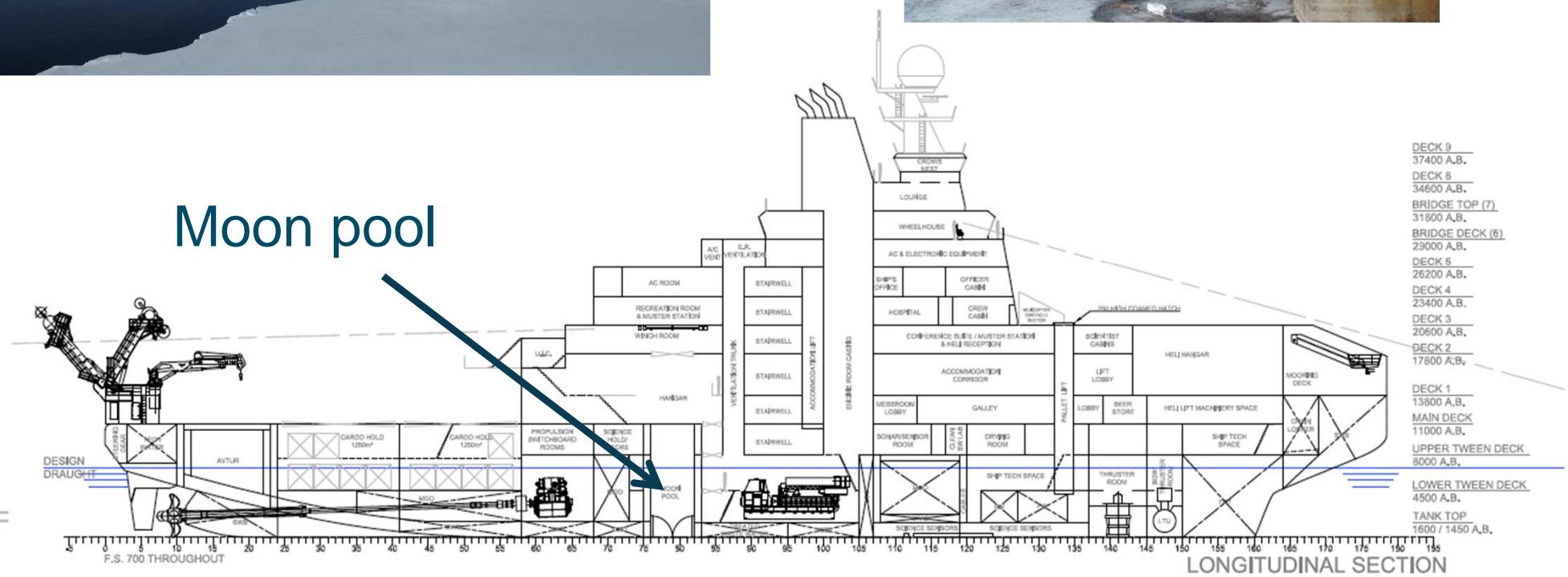
$$\Delta t = 16 \text{ d}$$

$$N(\text{O}_2) = (27 \pm 4) \text{ mmol m}^{-2} \text{ d}^{-1}$$

In-situ measurements can help avoid ship sampling biases



Intake design on research ships



Conclusions



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- ★ To understand variability on small to large scales, collaboration of all scientific disciplines is required.
- ★ Novel sensors and autonomous observation platforms will be key elements of future ocean biogeochemistry.
- ★ Biogeochemical data can supplement and sometimes substitute physical measurements.

- ★ Acknowledgements: present and former members of the Kaiser Lab; collaborators and colleagues at UEA, NOC, BAS, PML, CSIC, AWI, Princeton, Duke, etc.
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